

SPECIFICATION

NOVEL ARTIFICIAL NUCLEIC ACIDS OF
N-O BOND CROSSLINKAGE TYPE

5 TECHNICAL FIELD

[0001] This invention relates to oligonucleotide analogues, which have stable and excellent antisense or antigene activity or have excellent activities as reagents for detection of particular genes, or as primers for initiation of amplification, or which are useful as materials for various physiological and bioactive substances and pharmaceuticals, functional materials for double-stranded oligonucleotides for RNA interference method (RNAi or siRNA) or decoy method, functional materials for DNA chips or molecular beacons targeting single-stranded nucleic acids such as cDNA, functional materials for applications to various antisense methods (including ribozymes and DNAzymes), antigene method, and genetic homologous recombination, and materials for high sensitivity analysis of biological trace components by combinations with fluorescent or luminescent substances. The invention also relates to nucleoside analogues which are intermediates for production of the oligonucleotide analogues.

25 BACKGROUND ART

[0002] In 1978, it was reported for the first time that an antisense oligonucleotide (antisense molecule) inhibited infection by influenza virus. Reports followed that

antisense oligonucleotides also inhibited oncogene expression and AIDS infection. Since antisense oligonucleotides specifically control the expression of untoward genes, they are one of the fields that have been 5 expected most in recent years.

[0003] The antisense method is based on the concept that the flow of information in the so-called central dogma, i.e., DNA → mRNA → protein, is to be controlled using an antisense oligonucleotide.

10 [0004] However, if a natural DNA or RNA oligonucleotide was applied as an antisense molecule to this method, the problem arose that it was hydrolyzed by an in vivo enzyme, or its cell membrane permeation was not high. To resolve such problems, numerous nucleic acid derivative were 15 synthesized, and extensively studied. For example, phosphorothioates having an oxygen atom on the phosphorus atom substituted by a sulfur atom, and methyl phosphonates having a methyl group as a substituent were synthesized. Recently, products having the phosphorus atom also 20 substituted by a carbon atom, or molecules having a ribose converted into an acyclic skeleton form have been synthesized (F. Eckstein et al., Biochem., 18, 592(1979); P.S. Miller et al., Nucleic Acids Res., 11, 5189(1983); P. Herdewijn et al., J. Chem. Soc. Perkin Trans. 1, 25 1567(1993); P.E. Nielsen et al., Science, 254, 1497(1991)); C.A. Stein and A. M. Krieg (ed) "Applied Antisense Oligonucleotide Technology," Willy-Liss (1998); and J.J. Toulme et al., Prog. Nucl. Acid Rev. Mol. Biol., 67,

1(2001)).

[0005] However, none of the artificial oligonucleotides have obtained nucleoside and oligonucleotide analogues fully satisfactory in terms of a duplex-forming capacity 5 with single-stranded RNA and DNA, a triplex-forming capacity with double-stranded DNA, *in vivo* stability, or ease of synthesis of oligonucleotides.

DISCLOSURE OF THE INVENTION

PROBLEMS TO BE SOLVED BY THE INVENTION

10 [0006] In the light of the above-described conventional technologies, it is desired to provide a nucleotide analogue which highly penetrates the cell membrane *in vivo*, which minimally undergoes hydrolysis by enzymes, which is easy to synthesize, and which is useful for the antisense 15 method, the antigene method, the RNA interference method, the genetic homologous recombination method, and the decoy method.

MEANS FOR SOLVING THE PROBLEMS

[0007] The inventors of the present invention designed 20 a nucleic acid analogue having the sugar portion of a nucleic acid modified (i.e., artificial nucleic acid) which may be useful as a material for various physiological and bioactive substances and pharmaceuticals, a functional material for double-stranded oligonucleotides for the RNA 25 interference method (*Nature*, Vol. 411, 494-498, 2001) or the decoy method, a functional material for DNA chips or molecular beacons targeting single-stranded nucleic acids such as cDNA, a functional material for applications to

various antisense methods (including ribozymes and DNAzymes), the antigene method, and the genetic homologous recombination method, and a material for high sensitivity analysis of biological trace components by combinations 5 with fluorescent or luminescent substances. The inventors synthesized the nucleic acid analogue and confirmed its usefulness.

[0008] As will be described in detail by Examples and Experimental Examples to be offered later, a DNA or RNA 10 oligonucleotide analogue (II) containing an artificial nucleic acid 2',4'-BNA^{NC} unit [in the general formula (I), R₁ and R₂ are each hydrogen, and R₃ is hydrogen or a methyl group], which is one aspect of the present invention, was confirmed to have the following excellent characteristics:

15 (1) Has a very high duplex-forming capacity with a complementary RNA strand.

[0009] Whenever one 2',4'-BNA^{NC} unit is introduced into a DNA oligonucleotide (per modification), the Tm value rises by 3 to 6°C. Nevertheless, there is little increase 20 (improvement) in the duplex-forming capacity with a complementary DNA strand. In connection with this characteristic, a dramatic increase in the Tm value (marked improvement in the duplex-forming capacity) is present in binding affinity for a complementary RNA strand, as in the 25 case of a 2',4'-BNA-modified DNA oligonucleotide. On the other hand, the 2',4'-BNA-modified DNA oligonucleotide shows an improvement in the duplex-forming capacity with a complementary DNA strand (has the Tm value increased by 2

to 4°C per modification), as compared with an unmodified DNA oligonucleotide. By contrast, the above-mentioned 2',4'-BNA^{NC}-modified DNA oligonucleotide shows little improvement in binding affinity for a DNA strand. Thus,
5 this 2',4'-BNA^{NC}-modified DNA oligonucleotide is very superior in selective binding affinity for a RNA strand.

(2) The 2',4'-BNA^{NC}-modified DNA oligonucleotide also excels in a triplex-forming capacity with a double-stranded DNA chain.

10 [0010] When one 2',4'-BNA^{NC} unit is introduced into a DNA oligonucleotide, the T_m value rises by 7 to 12°C in forming a triplex with a double-stranded DNA chain. Triplex formation requires sequence selectivity such that base sequences should be strictly distinguished, and
15 binding only to a target sequence should be effected. The difference in the T_m value of the 2',4'-BNA^{NC}-modified DNA oligonucleotide with respect to a match sequence and a mismatch sequence is 25°C or higher. Thus, this oligonucleotide has excellent sequence selectivity
20 surpassing that of a natural DNA oligonucleotide.

(3) Nuclease resistance is unsurpassed.

[0011] The nuclease resistance of the 2',4'-BNA^{NC}-modified oligonucleotide is higher than that of the natural DNA oligonucleotide, but is much lower than that of S-oligo
25 (phosphorothioate type oligonucleotide). The 2',4'-BNA^{NC}-modified oligonucleotide of the present invention is superior in nuclease resistance not only to the 2',4'-BNA-modified oligonucleotide, but also to the S-oligo which has

been evaluated highly because of its excellent nuclease resistance. Thus, the 2',4'-BNA^{NC}-modified oligonucleotide of the present invention has the property of potently resisting degradation in vivo.

5 (4) The N-O bond contained in the artificial nucleic acid 2',4'-BNA^{NC} molecule of the present invention can be selectively cleaved under mild conditions with the use of a reducing reagent to liberate an NH group and an OH group. Binding of a different functional molecule, with the NH
10 group or the OH group as a foothold, makes it easy to obtain various conjugates, whether before or after preparation of the oligonucleotide analogue. Usable as the different functional group are labeling molecules such as fluorescent molecules, chemiluminescent molecules, and
15 molecular species containing radioisotope atoms, various DNA (RNA) incision activity molecules, and intracellular or nuclear transfer signal peptides.

[0012] As described above, the DNA or RNA oligonucleotide analogues of the present invention having 20 2',4'-BNA^{NC} modified in various forms have very high usefulness not only as highly functional materials for creation of genetic drugs by the antisense method, the antigene method, the decoy method, the genetic homologous recombination method, and the RNA interference method, but
25 also as base materials for genetic diagnosis methods such as molecular beacons and DNA chips, as well as starting materials for the development of research reagents for analysis and elucidation of gene functions.

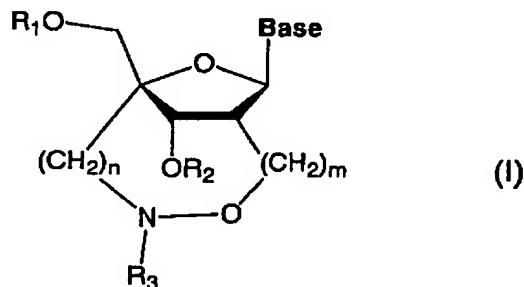
BRIEF DESCRIPTION OF THE DRAWING

[0013] [Fig. 1] A graph showing changes over time in an oligonucleotide analogue (8) of the present invention, and natural and nonnatural oligonucleotide analogues, when degraded by exonuclease, as the survival rates of the undegraded oligonucleotides by HPLC determination.

Embodiments of the Invention

[0014] The nucleoside analogues of the present invention are a compound of the following general formula 10 (I) and a salt thereof.

[0015] [Chemical Formula 1]



[0016] where Base represents an aromatic heterocyclic group or aromatic hydrocarbon ring group optionally having 15 a substituent,

[0017] R₁ and R₂ are identical or different, and each represent a hydrogen atom, a protective group for a hydroxyl group for nucleic acid synthesis, an alkyl group, an alkenyl group, a cycloalkyl group, an aryl group, an 20 aralkyl group, an acyl group, a sulfonyl group, a silyl group, a phosphate group, a phosphate group protected with a protective group for nucleic acid synthesis, or -P(R₄)R₅ [where R₄ and R₅ are identical or different, and each

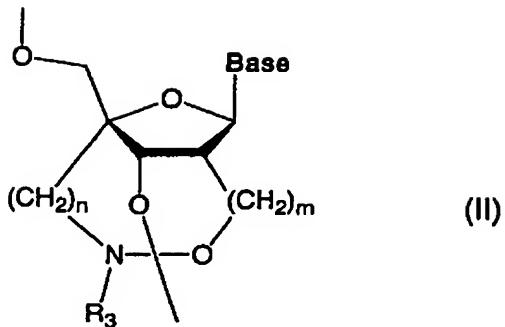
represent a hydroxyl group, a hydroxyl group protected with a protective group for nucleic acid synthesis, a mercapto group, a mercapto group protected with a protective group for nucleic acid synthesis, an amino group, an alkoxy group having 1 to 5 carbon atoms, an alkylthio group having 1 to 5 carbon atoms, a cyanoalkoxy group having 1 to 6 carbon atoms, or an amino group substituted by an alky group having 1 to 5 carbon atoms,

[0018] R_3 represents a hydrogen atom, an alkyl group, an alkenyl group, a cycloalkyl group, an aryl group, an aralkyl group, an acyl group, a sulfonyl group, or a functional molecule unit substituent, and

[0019] m denotes an integer of 0 to 2, and n denotes an integer of 1 to 3.

The oligonucleotide analogue of the present invention is a DNA oligonucleotide or RNA oligonucleotide analogue containing one or two or more of any one or more types of unit structures of nucleoside analogues represented by the following general formula (II), or a pharmacologically acceptable salt of the oligonucleotide analogue, provided that if two or more of one or more types of these structures are contained, Base may be identical or different between these structures, and that the bond between the respective nucleosides in the oligonucleotide analogue may contain one or two or more phosphorothioate bonds $[-OP(O)(S^-)O-]$ aside from a phosphodiester bond $[-OP(O_2^-)O-]$ identical with that in a natural nucleic acid,

[0020] [Chemical Formula 2]



[0021] where Base, R_1 , R_2 , R_3 , m, and n are as defined above.

In the general formulas (I) and (II), the aromatic heterocyclic group as Base refers to any 5-membered to 20-membered ring group which has a structure having the carbon atom, as a constituent atom of a hydrocarbon ring, substituted by one or more hetero-atoms such as nitrogen atoms, sulfur atoms or oxygen atoms, and which shows aromaticity, and includes a single ring or a condensed ring. Its concrete examples are pyrimidine or purine nucleic acid bases, and pyrimidine or purine nucleic acid bases optionally having one or more substituents selected from an α group to be described below. The pyrimidine or purine nucleic acid bases include bases generally known as constituent elements of nucleic acids (for example, guanine, adenine, cytosine, thymine, uracil), and all other chemical structures which are similar to them, and which can act as, or can be used instead of, the bases constituting nucleic acids. Other compounds are also included, such as thiophene, thianthrene, furan, pyran, isobenzofuran, chromene, xanthene, phenoxythine, pyrrole, imidazole, pyrazole, isothiazole, isoxazole, pyridazine, indolizine,

indole, isoindole, isoquinoline, quinoline, naphthyridine,
quinoxaline, quinazoline, pteridine, carbazole,
phenanthridine, acridine, perimidine, phenazine,
phenarsazine, phenothiazine, furazane, phenoxazine,
5 pyrrolidine, pyrroline, imidazolidine, imidazoline, and
pyrazolidine. Preferred examples are pyrimidine or purine
nucleic acid bases, and pyrimidine or purine nucleic acid
bases optionally having one or more substituents selected
from the α group to be described below. Concretely, a
10 purin-9-yl group, a 2-oxopyrimidin-1-yl group, or a purin-
9-yl group or a 2-oxopyrimidin-1-yl group having a
substituent selected from the following α group is
preferred.

[0022] α group: A hydroxyl group, a hydroxyl group
15 protected with a protective group for nucleic acid
synthesis, an alkoxy group having 1 to 5 carbon atoms, a
mercapto group, a mercapto group protected with a
protective group for nucleic acid synthesis, an alkylthio
group having 1 to 5 carbon atoms, an amino group, an amino
20 group protected with a protective group for nucleic acid
synthesis, an amino group substituted by an alkyl group
having 1 to 5 carbon atoms, an alkyl group having 1 to 5
carbon atoms, and a halogen atom. A group preferred as
the purine nucleic acid base optionally having the
25 substituent" is a 6-aminopurin-9-yl (i.e. adeninyl) group,
a 6-aminopurin-9-yl group having the amino group protected
with a protective group for nucleic acid synthesis, a 2,6-
diaminopurin-9-yl group, a 2-amino-6-chloropurin-9-yl group,

a 2-amino-6-chloropurin-9-yl group having the amino group protected with a protective group for nucleic acid synthesis, a 2-amino-6-fluoropurin-9-yl group, a 2-amino-6-fluoropurin-9-yl group having the amino group protected with a protective group for nucleic acid synthesis, a 2-amino-6-bromopurin-9-yl group, a 2-amino-6-bromopurin-9-yl group having the amino group protected with a protective group for nucleic acid synthesis, a 2-amino-6-hydroxypurin-9-yl (i.e., guaninyl) group, a 2-amino-6-hydroxypurin-9-yl group having the amino group protected with a protective group for nucleic acid synthesis, a 6-amino-2-methoxypurin-9-yl group, a 6-amino-2-chloropurin-9-yl group, a 6-amino-2-fluoropurin-9-yl group, a 2,6-dimethoxypurin-9-yl group, a 2,6-dichloroprolin-2-yl group, or a 6-mercaptopurin-9-yl group. More preferred is a 6-benzoylaminopurin-9-yl group, an adeninyl group, a 2-isobutyrylamino-6-hydroxypurin-9-yl group, or a guaninyl group.

[0023] A group preferred as "the pyrimidine nucleic acid base optionally having the substituent" is a 2-oxo-4-amino-1,2-dihydropyrimidin-1-yl (i.e., cytosinyl) group, a 2-oxo-4-amino-1,2-dihydropyrimidin-1-yl group having the amino group protected with a protective group for nucleic acid synthesis, a 2-oxo-4-amino-5-fluoro-1,2-dihydropyrimidin-1-yl group, a 2-oxo-4-amino-5-fluoro-1,2-dihydropyrimidin-1-yl group having the amino group protected with a protective group for nucleic acid synthesis, a 4-amino-2-oxo-5-chloro-1,2-dihydropyrimidin-1-yl group, a 2-oxo-4-methoxy-1,2-dihydropyrimidin-1-yl group,

a 2-oxo-4-mercaptop-1,2-dihydropyrimidin-1-yl group, a 2-oxo-4-hydroxy-1,2-dihydropyrimidin-1-yl (i.e., uracinyl) group, a 2-oxo-4-hydroxy-5-methyl-1,2-dihydropyrimidin-1-yl (i.e., thyminyl) group, or a 4-amino-5-methyl-2-oxo-1,2-dihydropyrimidin-1-yl (i.e., 5-methylcytosinyl) group.
5 More preferred is a 2-oxo-4-benzoylamino-1,2-dihydropyrimidin-1-yl group, a cytosinyl group, a thyminyl group, a uracinyl group, a 2-oxo-4-benzoylamino-5-methyl-1,2-dihydropyrimidin-1-yl group, or a 5-methylcytosinyl
10 group.

[0024] More preferred among "the purine or pyrimidine nucleic acid bases optionally having the substituent" is 6-aminopurin-9-yl (i.e. adeninyl), 6-aminopurin-9-yl having the amino group protected with a protective group for
15 nucleic acid synthesis, 2,6-diaminopurin-9-yl group, 2-amino-6-chloropurin-9-yl, 2-amino-6-chloropurin-9-yl having the amino group protected with a protective group for nucleic acid synthesis, 2-amino-6-fluoropurin-9-yl, 2-amino-6-fluoropurin-9-yl having the amino group protected
20 with a protective group for nucleic acid synthesis, 2-amino-6-bromopurin-9-yl, 2-amino-6-bromopurin-9-yl having the amino group protected with a protective group for nucleic acid synthesis, 2-amino-6-hydroxypurin-9-yl (i.e., guaninyl), 2-amino-6-hydroxypurin-9-yl having the amino
25 group protected with a protective group for nucleic acid synthesis, 6-amino-2-methoxypurin-9-yl, 6-amino-2-chloropurin-9-yl, 6-amino-2-fluoropurin-9-yl, 2,6-dimethoxypurin-9-yl, 2,6-dichloropurin-2-yl, 6-

mercaptopurin-9-yl, 2-oxo-4-amino-1,2-dihydropyrimidin-1-yl
(i.e., cytosinyl), 2-oxo-4-amino-1,2-dihydropyrimidin-1-yl
having the amino group protected with a protective group
for nucleic acid synthesis, 2-oxo-4-amino-5-fluoro-1,2-
5 dihydropyrimidin-1-yl, 2-oxo-4-amino-5-fluoro-1,2-
dihydropyrimidin-1-yl having the amino group protected with
a protective group for nucleic acid synthesis, 4-amino-
2-oxo-5-chloro-1,2-dihydropyrimidin-1-yl, 2-oxo-4-
methoxy-1,2-dihydropyrimidin-1-yl, 2-oxo-4-mercato-
10 1,2-dihydropyrimidin-1-yl, 2-oxo-4-hydroxy-1,2-
dihydropyrimidin-1-yl (i.e., uracinyl), 2-oxo-4-hydroxy-5-
methyl-1,2-dihydropyrimidin-1-yl (i.e., thyminyl),
4-amino-5-methyl-2-oxo-1,2-dihydropyrimidin-1-yl (i.e.,
5-methylcytosinyl), or 4-amino-5-methyl-2-oxo-1,2-
15 dihydropyrimidin-1-yl having the amino group protected with
a protective group for nucleic acid synthesis.

[0025] In the general formulas (I) and (II), the
aromatic hydrocarbon ring group as Base refers to a
monovalent substituent formed by removing one hydrogen atom
20 from a hydrocarbon ring having 6 to 20 carbon atoms and
showing aromaticity, and includes a single ring or a
condensed ring. Concrete examples are phenyl, indenyl,
naphthyl, pentalenyl, azulenyl, heptalenyl, biphenylenyl,
indacenyl, fluorenyl, phenanthryl, and anthryl. Any other
25 structures, which can be used alternatively as the base
portion of the nucleic acid component for the purpose of
the present invention, are also included in the examples.
Moreover, the aromatic hydrocarbon ring may be substituted

by one or more groups, such as a hydroxyl group, a hydroxyl group protected with a protective group for nucleic acid synthesis, an amino group, an amino group protected with a protective group for nucleic acid synthesis, a halogen atom,

5 a lower alkyl group, an alkoxy group, a carboxyl group, an aryloxy group, a nitro group, a trifluoromethyl group, and a phenyl group. Examples of such an optionally substituted aromatic hydrocarbon group are 4-hydroxyphenyl, 2-hydroxyphenyl, 4-aminophenyl, 2-aminophenyl, 2-methylphenyl,

10 2,6-dimethyphenyl, 2-chlorophenyl, 4-chlorophenyl, 2,4-dichlorophenyl, 2,5-dichlorophenyl, 2-bromophenyl, 4-methoxyphenyl, 4-chloro-2-nitrophenyl, 4-nitrophenyl, 2,4-dinitrophenyl, and biphenyl. Preferred examples of the optionally substituted aromatic hydrocarbon ring group are

15 a phenyl group, and a phenyl group substituted by a hydroxyl group, a hydroxyl group protected with a protective group for nucleic acid synthesis, an amino group, an amino group protected with a protective group for nucleic acid synthesis, a lower alkoxy group, or a nitro group.

20

[0026] In the general formula (I) or (II), the "protective group for a hydroxyl group for nucleic acid synthesis" as R₁ and R₂, and the protective group in the "hydroxyl group protected with a protective group for nucleic acid synthesis" as R₄ and R₅ and in the α group are not limited, as long as they can protect the hydroxyl group stably during nucleic acid synthesis. Concretely, they refer to protective groups which are stable under acidic or

neutral conditions, and which can be cleaved by a chemical method such as hydrogenolysis, hydrolysis, electrolysis, or photolysis. Examples of such protective groups are "aliphatic acyl groups", for example, alkylcarbonyl groups such as formyl, acetyl, propionyl, butyryl, isobutyryl, pentanoyl, pivaloyl, valeryl, isovaleryl, octanoyl, nonanoyl, decanoyl, 3-methylnonanoyl, 8-methylnonanoyl, 3-ethyloctanoyl, 3,7-dimethyloctanoyl, undecanoyl, dodecanoyl, tridecanoyl, tetradecanoyl, pentadecanoyl, hexadecanoyl, 1-methylpentadecanoyl, 14-methylpentadecanoyl, 13,13-dimethyltetradecanoyl, heptadecanoyl, 15-methylhexadecanoyl, octadecanoyl, 1-methylheptadecanoyl, nonadecanoyl, eicosanoyl, and heneicosanoyl, carboxylated alkylcarbonyl groups such as succinoyl, glutaroyl, and adipoyl, halogeno lower alkylcarbonyl groups such as chloroacetyl, dichloroacetyl, trichloroacetyl, and trifluoroacetyl, lower alkoxy lower alkylcarbonyl groups such as methoxyacetyl, and unsaturated alkylcarbonyl groups such as (E)-2-methyl-2-butenoyl; "lower alkyl groups" such as methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, s-butyl, tert-butyl, n-pentyl, isopentyl, 2-methylbutyl, neopentyl, 1-ethylpropyl, n-hexyl, isoheptyl, 4-methylpentyl, 3-methylpentyl, 2-methylpentyl, 1-methylpentyl, 3,3,-dimethylbutyl, 2,2-dimethylbutyl, 1,1-dimethylbutyl, 1,2-dimethylbutyl, 1,3-dimethylbutyl, 2,3-dimethylbutyl, and 2-ethylbutyl; "lower alkenyl groups" such as ethenyl, 1-propenyl, 2-propenyl, 1-methyl-2-propenyl, 1-methyl-1-propenyl, 2-methyl-1-propenyl, 2-methyl-2-propenyl, 2-

ethyl-2-propenyl, 1-butenyl, 2-butenyl, 1-methyl-2-butenyl,
1-methyl-1-butenyl, 3-methyl-2-butenyl, 1-ethyl-2-butenyl,
3-butenyl, 1-methyl-3-butenyl, 2-methyl-3-butenyl, 1-ethyl-
3-butenyl, 1-pentenyl, 2-pentenyl, 1-methyl-2-pentenyl, 2-
5 methyl-2-pentenyl, 3-pentenyl, 1-methyl-3-pentenyl, 2-
methyl-3-pentenyl, 4-pentenyl, 1-methyl-4-pentenyl, 2-
methyl-4-pentenyl, 1-hexenyl, 2-hexenyl, 3-hexenyl, 4-
hexenyl, and 5-hexenyl; "aromatic acyl groups", for example,
arylcarbonyl groups such as benzoyl, α -naphthoyl, and β -
10 naphthoyl, halogenoarylcarbonyl groups such as 2-
bromobenzoyl and 4-chlorobenzoyl, lower alkylated
arylcarbonyl groups such as 2,4,6-trimethylbenzoyl and 4-
toluoyl, lower alkoxylated arylcarbonyl groups such as 4-
anisoyl, carboxylated arylcarbonyl groups such as 2-
15 carboxybenzoyl, 3-carboxybenzoyl, and 4-carboxybenzoyl,
nitrated arylcarbonyl groups such as 4-nitrobenzoyl and 2-
nitrobenzoyl, lower alkoxy carbonylated arylcarbonyl groups
such as 2-(methoxycarbonyl)benzoyl, and arylated
arylcarbonyl groups such as 4-phenylbenzoyl;
20 "tetrahydropyranyl or tetrahydrothiopyranyl groups" such as
tetrahydropyran-2-yl, 3-bromotetrahydropyran-2-yl, 4-
methoxytetrahydropyran-4-yl, tetrahydrothiopyran-4-yl, and
4-methoxytetrahydrothiopyran-4-yl; "tetrahydrofuranyl or
tetrahydrothiofuranyl groups" such as tetrahydrofuran-2-yl
25 and tetrahydrothiofuran-2-yl; "silyl groups", for example,
lower trialkylsilyl groups such as trimethylsilyl,
triethylsilyl, isopropyldimethylsilyl, t-butyldimethylsilyl,
methyldiisopropylsilyl, methyldi-t-butylsilyl, and

triisopropylsilyl, lower alkylsilyl groups substituted by one or two aryl groups, such as diphenylmethysilyl, diphenylbutylsilyl, diphenylisopropylsilyl, and phenyldiisopropylsilyl; "lower alkoxymethyl groups" such as 5 methoxymethyl, 1,1-dimethyl-1-methoxymethyl, ethoxymethyl, propoxymethyl, isopropoxymethyl, butoxymethyl, and tert-butoxymethyl; "lower alkoxylated lower alkoxymethyl groups" such as 2-methoxyethoxymethyl; "halogeno lower alkoxymethyl groups" such as 2,2,2-trichloroethoxymethyl and bis(2-chloroethoxy)methyl; "lower alkoxylated ethyl groups" such as 10 1-ethoxyethyl and 1-(isopropoxy)ethyl; "halogenated ethyl groups" such as 2,2,2-trichloroethyl; "methyl groups substituted by one to three aryl groups" such as benzyl, α-naphthylmethyl, β-naphthylmethyl, diphenylmethyl, triphenylmethyl, α-naphthyldiphenylmethyl, and 9-anthrylmethyl; "methyl groups substituted by one to three aryl groups having an aryl ring substituted by a lower 15 alkyl, lower alkoxy, halogen, or cyano group", such as 4-methylbenzyl, 2,4,6-trimethylbenzyl, 3,4,5-trimethylbenzyl, 4-methoxybenzyl, 4-methoxyphenyldiphenylmethyl, 4,4'-dimethoxytriphenylmethyl, 2-nitrobenzyl, 4-nitrobenzyl, 4-chlorobenzyl, 4-bromobenzyl, and 4-cyanobenzyl; "lower 20 alkoxycarbonyl groups" such as methoxycarbonyl, ethoxycarbonyl, t-butoxycarbonyl, and isobutoxycarbonyl; "aryl groups substituted by a halogen atom, a lower alkoxy group, or a nitro group", such as 4-chlorophenyl, 2-fluorophenyl, 4-methoxyphenyl, 4-nitrophenyl, and 2,4-dinitrophenyl; "lower alkoxycarbonyl groups substituted by 25

halogen or a lower trialkylsilyl group", such as 2,2,2-trichloroethoxycarbonyl and 2-trimethylsilylethoxycarbonyl; "alkenyloxycarbonyl groups" such as vinyloxycarbonyl and aryloxycarbonyl; and "aralkyloxycarbonyl groups having an 5 aryl ring optionally substituted by one or two lower alkoxy or nitro groups", such as benzyloxycarbonyl, 4-methoxybenzyloxycarbonyl, 3,4-dimethoxybenzyloxycarbonyl, 2-nitrobenzyloxycarbonyl, and 4-nitrobenzyloxycarbonyl.

[0027] The "protective group for a hydroxyl group for 10 nucleic acid synthesis", as R₁ and R₂, is preferably the "aliphatic acyl group", the "aromatic acyl group", the "methyl group substituted by one to three aryl groups", the "methyl group substituted by one to three aryl groups having an aryl ring substituted by a lower alkyl, lower 15 alkoxy, halogen, or cyano group", or the "silyl group", more preferably an acetyl group, a benzoyl group, a benzyl group, a p-methoxybenzoyl group, a dimethoxytrityl group, a monomethoxytrityl group, or a tert-butyldiphenylsilyl group. The protective group in the "hydroxyl group protected with 20 a protective group for nucleic acid synthesis", as R₄, R₅, R₆, and R₇, and in the α group, is preferably the "aliphatic acyl group", the "aromatic acyl group", the "methyl group substituted by one to three aryl groups", the "aryl group substituted by a halogen atom, a lower alkoxy group, or a 25 nitro group", the "lower alkyl group", or the "lower alkenyl group", more preferably a benzoyl group, a benzyl group, a 2-chlorophenyl group, a 4-chlorophenyl group, or a 2-propenyl group.

[0028] In the general formula (I) or (II), the "alkyl group", as R₁, R₂ and R₃, refers to a straight chain or branched chain alkyl group having 1 to 20 carbon atoms, and includes a straight chain or branched chain alkyl group
5 having 1 to 6 carbon atoms (such an alkyl group may herein be referred to as a lower alkyl group), such as methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, s-butyl, tert-butyl, n-pentyl, isopentyl, 2-methylbutyl, neopentyl,
10 1-ethylpropyl, n-hexyl, isohexyl, 4-methylpentyl, 3-methylpentyl, 2-methylpentyl, 1-methylpentyl, 3,3,-dimethylbutyl, 2,2-dimethylbutyl, 1,1-dimethylbutyl, 1,2-dimethylbutyl, 1,3-dimethylbutyl, 2,3-dimethylbutyl, or 2-ethylbutyl. The alkyl group also includes a straight chain or branched chain alkyl group having 7 to 20 carbon atoms,
15 such as heptyl, octyl, nonyl or decyl. Preferred is the above-mentioned straight chain or branched chain alkyl group having 1 to 6 carbon atoms.

[0029] In the general formula (I) or (II), the "alkenyl group", as R₁, R₂ and R₃, refers to a straight chain or branched chain alkenyl group having 2 to 20 carbon atoms, and includes a straight chain or branched chain alkenyl group having 2 to 6 carbon atoms (such an alkenyl group may herein be referred to as a lower alkenyl group), such as ethenyl, 1-propenyl, 2-propenyl, 1-methyl-2-propenyl, 1-methyl-1-propenyl, 2-methyl-1-propenyl, 2-methyl-2-propenyl, 2-ethyl-2-propenyl, 1-butenyl, 2-butenyl, 1-methyl-2-butenyl, 1-methyl-1-butenyl, 3-methyl-2-butenyl, 1-ethyl-2-butenyl, 3-butenyl, 1-methyl-3-butenyl, 2-methyl-3-butenyl,

1-ethyl-3-butenyl, 1-pentenyl, 2-pentenyl, 1-methyl-2-pentenyl, 2-methyl-2-pentenyl, 3-pentenyl, 1-methyl-3-pentenyl, 2-methyl-3-pentenyl, 4-pentenyl, 1-methyl-4-pentenyl, 2-methyl-4-pentenyl, 1-hexenyl, 2-hexenyl, 3-hexenyl, 4-hexenyl, and 5-hexenyl. The alkeyl group also includes geranyl and farnesyl, and is preferably the above-mentioned straight chain or branched chain alkenyl group having 2 to 6 carbon atoms.

[0030] In the general formula (I) or (II), the "cycloalkyl group", as R_1 , R_2 and R_3 , refers to a cycloalkyl group having 3 to 10 carbon atoms, and includes, for example, cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cycloheptyl, cyclooctyl, norbornyl, and adamantyl. Preferred is a cycloalkyl group having 3 to 8 carbon atoms, such as cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cycloheptyl, or cyclooctyl. The "cycloalkyl group" also includes a heterocyclic group in which one or more methylene groups on the ring of the cycloalkyl group have been substituted by oxygen atoms or sulfur atoms, or nitrogen atoms substituted by an alkyl group. An example of the heterocyclic group is a tetrahydropyranyl group.

[0031] In the general formula (I) or (II), the "aryl group", as R_1 , R_2 and R_3 , refers to a monovalent substituent having 6 to 14 carbon atoms which remains after removing one hydrogen atom from an aromatic hydrocarbon group, and includes, for example, phenyl, indenyl, naphthyl, phenanthrenyl, and anthracenyl. The aryl group may be substituted by one or more groups, such as a halogen atom,

a lower alkyl group, a hydroxyl group, an alkoxy group, an aryloxy group, an amino group, a nitro group, trifluoromethyl, and a phenyl group. Examples of the optionally substituted aryl group are 2-methylphenyl, 2,6-
5 dimethylphenyl, 2-chlorophenyl, 4-chlorophenyl, 2,4-dichlorophenyl, 2,5-dichlorophenyl, 2-bromophenyl, 4-methoxyphenyl, 4-chloro-2-nitrophenyl, 4-nitrophenyl, 2,4-dinitrophenyl, and biphenyl. Preferred examples are a phenyl group, and a phenyl group substituted by a halogen
10 atom, a lower alkoxy group, or a nitro group.

[0032] In the general formula (I) or (II), the "aralkyl group", as R₁, R₂ and R₃, refers to an alkyl group having 1 to 6 carbon atoms which has been substituted by an aryl group. Examples of the aralkyl group are "methyl groups substituted by one to three aryl groups", such as benzyl, α-naphthylmethyl, β-naphthylmethyl, indenylmethyl, phenanthrenylmethyl, anthracenylmethyl, diphenylmethyl, triphenylmethyl, α-naphthyldiphenylmethyl, and 9-anthrylmethyl, and "methyl groups substituted by one to three aryl groups having an aryl ring substituted by a lower alkyl, lower alkoxy, halogen, or cyano group", such as 4-methylbenzyl, 2,4,6-trimethylbenzyl, 3,4,5-trimethylbenzyl, 4-methoxybenzyl, 4-methoxyphenyldiphenylmethyl, 4,4'-dimethoxytriphenylmethyl, 25 2-nitrobenzyl, 4-nitrobenzyl, 4-chlorobenzyl, 4-bromobenzyl, and 4-cyanobenzyl. Other examples include "alkyl groups having 3 to 6 carbon atoms substituted by an aryl group", such as 1-phenethyl, 2-phenethyl, 1-naphthylethyl, 2-

naphthylethyl, 1-phenylpropyl, 2-phenylpropyl, 3-phenylpropyl, 1-naphthylpropyl, 2-naphthylpropyl, 3-naphthylpropyl, 1-phenylbutyl, 2-phenylbutyl, 3-phenylbutyl, 4-phenylbutyl, 1-naphthylbutyl, 2-naphthylbutyl, 3-naphthylbutyl, 4-naphthylbutyl, 1-phenylpentyl, 2-phenylpentyl, 3-phenylpentyl, 4-phenylpentyl, 5-phenylpentyl, 1-naphthylpentyl, 2-naphthylpentyl, 3-naphthylpentyl, 4-naphthylpentyl, 5-naphthylpentyl, 1-phenylhexyl, 2-phenylhexyl, 3-phenylhexyl, 4-phenylhexyl, 5-phenylhexyl, 6-phenylhexyl, 1-naphthylpentyl, 2-naphthylpentyl, 3-naphthylpentyl, 4-naphthylpentyl, 5-naphthylpentyl, and 6-naphthylpentyl. Preferred examples are the "methyl groups substituted by one to three aryl groups", and the "methyl groups substituted by one to three aryl groups having an aryl ring substituted by a lower alkyl, lower alkoxy, halogen, or cyano group". More preferred examples are 4-methoxyphenyldiphenylmethyl, and 4,4'-dimethoxytriphenylmethyl.

[0033] Examples of the "acryl group", as R_1 , R_2 and R_3 , in the general formula (I) or (II) are "aliphatic acyl groups", for example, alkylcarbonyl groups such as formyl, acetyl, propionyl, butyryl, isobutyryl, pentanoyl, pivaloyl, valeryl, isovaleryl, octanoyl, nonanoyl, decanoyl, 3-methylnonanoyl, 8-methylnonanoyl, 3-ethyloctanoyl, 3,7-dimethyloctanoyl, undecanoyl, dodecanoyl, tridecanoyl, tetradecanoyl, pentadecanoyl, hexadecanoyl, 1-methylpentadecanoyl, 14-methylpentadecanoyl, 13,13-dimethyltetradecanoyl, heptadecanoyl, 15-methylhexadecanoyl,

octadecanoyl, 1-methylheptadecanoyl, nonadecanoyl, eicosanoyl, and heneicosanoyl, carboxylated alkylcarbonyl groups such as succinoyl, glutaroyl, and adipoyl, halogeno lower alkylcarbonyl groups such as chloroacetyl,

5 dichloroacetyl, trichloroacetyl, and trifluoroacetyl, lower alkoxy lower alkylcarbonyl groups such as methoxyacetyl, aryloxy lower alkylcarbonyl groups such as a phenoxyacetyl group, and unsaturated alkylcarbonyl groups such as (E)-2-methyl-2-butenoyl; and "aromatic acyl groups", for example,

10 arylcarbonyl groups such as benzoyl, α -naphthoyl, and β -naphthoyl, halogenoarylcarbonyl groups such as 2-bromobenzoyl and 4-chlorobenzoyl, lower alkylated arylcarbonyl groups such as 2,4,6-trimethylbenzoyl and 4-toluoyl, lower alkoxylated arylcarbonyl groups such as

15 4-anisoyl, carboxylated arylcarbonyl groups such as 2-carboxybenzoyl, 3-carboxybenzoyl, and 4-carboxybenzoyl, nitrated arylcarbonyl groups such as 4-nitrobenzoyl and 2-nitrobenzoyl, lower alkoxycarbonylated arylcarbonyl groups such as 2-(methoxycarbonyl)benzoyl, and arylated

20 arylcarbonyl groups such as 4-phenylbenzoyl. Preferred examples are formyl, acetyl, propionyl, butyryl, isobutyryl, pentanoyl, pivaloyl, benzoyl, and phenoxyacetyl groups.

[0034] As the "sulfonyl group", as R_1 , R_2 and R_3 , in the general formula (I) or (II), there can be named "aliphatic

25 sulfonyl groups", for example, sulfonyl groups substituted by a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, such as methanesulfonyl and ethanesulfonyl, and "aromatic sulfonyl groups", for example,

sulfonyl groups substituted by various aryl groups, such as benzenesulfonyl and p-toluenesulfonyl. Preferred examples are methanesulfonyl and p-toluenesulfonyl.

[0035] As the "silyl group", as R₁, R₂ and R₃, in the general formula (I) or (II), there can be named "lower trialkylsilyl groups" such as trimethylsilyl, triethylsilyl, isopropyldimethylsilyl, tert-butyldimethylsilyl, methyldiisopropylsilyl, methyldi-tert-butylsilyl, and triisopropylsilyl, "lower alkylsilyl groups substituted by one or two aryl groups", such as diphenylmethysilyl, tert-butyldiphenylbutylsilyl, diphenylisopropylsilyl, and phenyldiisopropylsilyl. Preferred examples are trimethylsilyl, triethylsilyl, triisopropylsilyl, t-butyldimethylsilyl, t-butyldiphenylsilyl. A more preferred example is trimethylsilyl.

[0036] In the general formula (I) or (II), the "protective group" in the "phosphate group protected with a protective group for nucleic acid synthesis" as R₁ and R₂ is not limited, as long as it can protect the phosphate group stably during nucleic acid synthesis. Concretely, it refers to a protective group which is stable under acidic or neutral conditions, and which can be cleaved by a chemical method such as hydrogenolysis, hydrolysis, electrolysis, or photolysis. Examples of such a protective group are "lower alkyl groups" such as methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, s-butyl, tert-butyl, n-pentyl, isopentyl, 2-methylbutyl, neopentyl, 1-ethylpropyl, n-hexyl, isohexyl, 4-methylpentyl, 3-

methylpentyl, 2-methylpentyl, 1-methylpentyl, 3,3,-
dimethylbutyl, 2,2-dimethylbutyl, 1,1-dimethylbutyl, 1,2-
dimethylbutyl, 1,3-dimethylbutyl, 2,3-dimethylbutyl, and
2-ethylbutyl; "cyanated lower alkyl groups" such as 2-
5 cyanoethyl, and 2-cyano-1,1-dimethylethyl; "ethyl groups
substituted by a silyl group", such as 2-
methyldiphenylsilylethyl, 2-trimethylsilylethyl, and 2-
triphenylsilylethyl; "halogenated lower alkyl groups", such
as 2,2,2-trichloroethyl, 2,2,2-tribromoethyl, 2,2,2-
10 trifluoroethyl, and 2,2,2-trichloro-1,1-diemthylethyl;
"lower alkenyl groups", such as ethenyl, 1-propenyl, 2-
propenyl, 1-methyl-2-propenyl, 1-methyl-1-propenyl, 2-
methyl-2-propenyl, 2-ethyl-2-propenyl, 1-butenyl, 2-butenyl,
1-methyl-2-butenyl, 1-methyl-1-butenyl, 3-methyl-2-butenyl,
15 1-ethyl-2-butenyl, 3-butenyl, 1-methyl-3-butenyl, 2-methyl-
3-butenyl, 1-ethyl-3-butenyl, 1-pentenyl, 2-pentenyl, 1-
methyl-2-pentenyl, 2-methyl-2-pentenyl, 3-pentenyl, 1-
methyl-3-pentenyl, 2-methyl-3-pentenyl, 4-pentenyl, 1-
methyl-4-pentenyl, 2-methyl-4-pentenyl, 1-hexenyl, 2-
20 hexenyl, 3-hexenyl, 4-hexenyl, and 5-hexenyl; "cycloalkyl
groups", such as cyclopropyl, cyclobutyl, cyclopentyl,
cyclohexyl, cycloheptyl, norbornyl, and adamantlyl;
"cyanated lower alkenyl groups" such as 2-cyanobutenyl;
"aralkyl groups", such as benzyl, α -naphthylmethyl,
25 β -naphthylmethyl, indenylmethyl, phenanthrenylmethyl,
anthracenylmethyl, diphenylmethyl, triphenylmethyl, 1-
phenethyl, 2-phenethyl, 1-naphthylethyl, 2-naphthylethyl,
1-phenylpropyl, 2-phenylpropyl, 3-phenylpropyl, 1-

naphthylpropyl, 2-naphthylpropyl, 3-naphthylpropyl, 1-phenylbutyl, 2-phenylbutyl, 3-phenylbutyl, 4-phenylbutyl, 1-naphthylbutyl, 2-naphthylbutyl, 3-naphthylbutyl, 4-naphthylbutyl, 1-phenylpentyl, 2-phenylpentyl, 3-phenylpentyl, 4-phenylpentyl, 5-phenylpentyl, 1-naphthylpentyl, 2-naphthylpentyl, 3-naphthylpentyl, 4-naphthylpentyl, 5-naphthylpentyl, 1-phenylhexyl, 2-phenylhexyl, 3-phenylhexyl, 4-phenylhexyl, 5-phenylhexyl, 6-phenylhexyl, 1-naphthylpentyl, 2-naphthylpentyl, 3-naphthylpentyl, 4-naphthylpentyl, 5-naphthylpentyl, and 6-naphthylpentyl; "aralkyl groups having an aryl ring substituted by a nitro group or a halogen atom", such as 4-chlorobenzyl, 2-(4-nitrophenyl)ethyl, o-nitrobenzyl, 4-nitrobenzyl, 2,4-dinitrobenzyl, and 4-chloro-2-nitrobenzyl; "aryl groups", such as phenyl, indenyl, naphthyl, phenanthrenyl, and anthracenyl; and "aryl groups substituted by a lower alkyl group, a halogen atom, or a nitro group", such as 2-methylphenyl, 2,6-dimethylphenyl, 2-chlorophenyl, 4-chlorophenyl, 2,4-dichlorophenyl, 2,5-dichlorophenyl, 2-bromophenyl, 4-nitrophenyl, and 4-chloro-2-nitrophenyl. Preferred examples are the "lower alkyl groups", "lower alkyl groups substituted by a cyano group", "aralkyl groups", "aralkyl groups having an aryl ring substituted by a nitro group or a halogen atom", or "aryl groups substituted by a lower alkyl group, a halogen atom, or a nitro group". A more preferred example is a 2-cyanoethyl group, a 2,2,2-trichloroethyl group, a benzyl group, a 2-chlorophenyl group, or a 4-chlorophenyl group.

[0037] In the general formula (I) or (II), the "functional molecule unit substituent" as R₃ includes labeling molecules (for example, fluorescent molecules, chemiluminescent molecules, and molecular species containing radioisotope atoms), DNA or RNA incision activity molecules, and intracellular or nuclear transfer signal peptides.

[0038] In the general formula (I) or (II), the protective group in "the mercapto group protected with a protective group for nucleic acid synthesis" as R₄ and R₅ and in the α group is not limited, as long as it can protect the mercapto group stably during nucleic acid synthesis. Concretely, it refers to a protective group which is stable under acidic or neutral conditions, and which can be cleaved by a chemical method such as hydrogenolysis, hydrolysis, electrolysis, or photolysis. Examples of such a protective group are those named above as the protective group for the hydroxyl group, as well as "disulfide-forming groups", for example, alkylthio groups such as methylthio, ethylthio, and tert-butylthio, and arylthio groups such as benzylthio. Preferred examples are "aliphatic acyl groups" or "aromatic acyl groups". More preferred examples are a benzoyl group and a benzyl group.

[0039] Examples of "the alkoxy group having 1 to 5 carbon atoms" as R₄ and R₅ and in the α group in the general formula (I) or (II) are methoxy, ethoxy, n-propoxy, isopropoxy, n-butoxy, isobutoxy, s-butoxy, tert-butoxy, and n-pentoxy. A preferred example is a methoxy or ethoxy

group.

[0040] Examples of "the alkylthio group having 1 to 5 carbon atoms" as R₄ and R₅ and in the α group in the general formula (I) or (II) are methylthio, ethylthio, propylthio, 5 isopropylthio, butylthio, isobutylthio, s-butylthio, tert-butylthio, and n-pentylthio. A preferred example is a methylthio or ethylthio group.

[0041] Examples of "the cyanoalkoxy group having 1 to 6 carbon atoms" as R₄ and R₅ in the general formula (I) or 10 (II) are the above "alkoxy groups having 1 to 5 carbon atoms" which have been substituted by a cyano group. Such groups are, for example, cyanomethoxy, 2-cyanoethoxy, 3-cyanopropoxy, 4-cyanobutoxy, 3-cyano-2-methylpropoxy, and 15 1-cyanomethyl-1,1-dimethylmethoxy. A preferred example is a 2-cyanoethoxy group.

[0042] Examples of "the amino group substituted by an alkyl group having 1 to 5 carbon atoms", as R₄ and R₅ and in the α group in the general formula (I) or (II), are 20 methylamino, ethylamino, propylamino, isopropylamino, butylamino, isobutylamino, s-butylamino, tert-butylamino, dimethylamino, diethylamino, dipropylamino, diisopropylamino, dibutylamino, diisobutylamino, di(s-butyl)amino, and di(tert-butyl)amino. A preferred example 25 is methylamino, ethylamino, dimethylamino, diethylamino, or diisopropylamino group.

[0043] As the "alkyl group having 1 to 5 carbon atoms" in the α group, there can be named, for example, methyl, ethyl, propyl, isopropyl, isopropyl, butyl, isobutyl,

s-butyl, tert-butyl, and n-pentyl. A preferred example is a methyl or ethyl group.

[0044] As the "halogen atom" in the α group, there can be named, for example, a fluorine atom, a chlorine atom, a 5 bromine atom, and an iodine atom, A preferred example is a fluorine atom or a chlorine atom.

[0045] The "phosphoroamidite group" refers to a group of the formula $-P(OR_{1a})(NR_{1b})$ (where R_{1a} represents an alkyl group having 1 or more carbon atoms or a cyanoalkyl group 10 having 1 to 7 carbon atoms, and R_{1b} represents an alkyl group having 1 to 6 carbon atoms). A preferred example is a group of the formula $-P(OC_2H_4CN)(N(i-Pr)_2)$ or a group of the formula $-P(OCH_3)(N(i-Pr)_2)$.

[0046] The protective group in the "amino group 15 protected with a protective group for nucleic acid synthesis" in the α group is not limited, as long as it can protect the amino group stably during nucleic acid synthesis. Concretely, it refers to a protective group which is stable under acidic or neutral conditions, and 20 which can be cleaved by a chemical method such as hydrogenolysis, hydrolysis, electrolysis, or photolysis. Examples of such a protective group are "aliphatic acyl groups", for example, alkylcarbonyl groups such as formyl, acetyl, propionyl, butyryl, isobutyryl, pentanoyl, pivaloyl, 25 valeryl, isovaleryl, octanoyl, nonanoyl, decanoyl, 3-methylnonanoyl, 8-methylnonanoyl, 3-ethyloctanoyl, 3,7-dimethyloctanoyl, undecanoyl, dodecanoyl, tridecanoyl, tetradecanoyl, pentadecanoyl, hexadecanoyl, 1-

methylpentadecanoyl, 14-methylpentadecanoyl, 13,13-dimethyltetradecanoyl, heptadecanoyl, 15-methylhexadecanoyl, octadecanoyl, 1-methylheptadecanoyl, nonadecanoyl, nonadecanoyl, eicosanoyl, and heneicosanoyl, carboxylated alkylcarbonyl groups such as succinoyl, glutaroyl, and adipoyl, halogeno lower alkylcarbonyl groups such as chloroacetyl, dichloroacetyl, trichloroacetyl, and trifluoroacetyl, lower alkoxy lower alkylcarbonyl groups such as methoxyacetyl, and unsaturated alkylcarbonyl groups such as (E)-2-methyl-2-butenoyl; "aromatic acyl groups", for example, arylcarbonyl groups such as benzoyl, α-naphthoyl, and β-naphthoyl, halogenoarylcarbonyl groups such as 2-bromobenzoyl and 4-chlorobenzoyl, lower alkylated arylcarbonyl groups such as 2,4,6-trimethylbenzoyl and 4-toluoyl, lower alkoxylated arylcarbonyl groups such as 4-anisoyl, carboxylated arylcarbonyl groups such as 2-carboxybenzoyl, 3-carboxybenzoyl, and 4-carboxybenzoyl, nitrated arylcarbonyl groups such as 4-nitrobenzoyl and 2-nitrobenzoyl, lower alkoxycarbonylated arylcarbonyl groups such as 2-(methoxycarbonyl)benzoyl, and arylated arylcarbonyl groups such as 4-phenylbenzoyl; "lower alkoxycarbonyl groups" such as methoxycarbonyl, ethoxycarbonyl, t-butoxycarbonyl, and isobutoxycarbonyl; "lower alkoxycarbonyl groups substituted by halogen or a lower trialkylsilyl group", such as 2,2,2-trichloroethoxycarbonyl and 2-trimethylsilylethoxycarbonyl; "alkenyloxycarbonyl groups" such as vinyloxycarbonyl and aryloxycarbonyl; and "aralkyloxycarbonyl groups having an

aryl ring optionally substituted by one or two lower alkoxy or nitro groups", such as benzyloxycarbonyl, 4-methoxybenzyloxycarbonyl, 2-nitrobenzyloxycarbonyl, and 4-nitrobenzyloxycarbonyl. A preferred example is the
5 "aliphatic acyl group" or "aromatic acyl group", and a more preferred example is a benzoyl group.

[0047] The "nucleoside analogue" refers to a nonnatural type of "nucleoside" consisting of a purine or pyrimidine base and a sugar linked together, or a product consisting
10 of an aromatic heterocyclic ring or an aromatic hydrocarbon ring, which is other than purine and pyrimidine and which can be used instead of the purine or pyrimidine base, and a sugar linked together.

[0048] The "oligonucleotide analogue" refers to a nonnatural type derivative of "oligonucleotide" comprising
15 2 to 50 identical or different "nucleosides" or "nucleoside analogues" linked together by phosphodiester bonds.
Preferred examples of such an analogue are sugar derivatives with the sugar portion modified; thioate
20 derivatives formed upon thioation of the phosphodiester portion; esters formed upon esterification of the terminal phosphoric acid portion; and amides formed upon amidation of the amino group on the purine base. More preferred examples are sugar derivatives with the sugar portion
25 modified.

[0049] The "salt thereof" refers to a salt of the compound (1) of the present invention, because the compound (1) can be converted into the salt. Preferred examples of

the salt are metal salts, for example, alkali metal salts such as sodium salt, potassium salt, and lithium salt, alkaline earth metal salts such as calcium salt and magnesium salt, aluminum salt, iron salt, zinc salt, copper 5 salt, nickel salt, and cobalt salt; amine salts, for example, inorganic salts such as ammonium salt, and organic salts such as t-octylamine salt, dibenzylamine salt, morpholine salt, glucosamine salt, phenylglycine alkyl ester salt, ethylenediamine salt, N-methylglucamine salt, 10 guanidine salt, diethylamine salt, triethylamine salt, dicyclohexylamine salt, N,N'-dibenzylethylenediamine salt, chloroprocaine salt, procaine salt, diethanolamine salt, N-benzyl-phenethylamine salt, piperazine salt, tetramethylammonium salt, and 15 tris(hydroxymethyl)aminomethane salt; inorganic acid salts, for example, halogenated hydroacid salts such as hydrofluoride, hydrochloride, hydrobromide, and hydriodide, nitrate, perchlorate, sulfate, and phosphate; organic acid salts, for example, lower alkanesulfonates such as 20 methanesulfonate, trifluoromethanesulfonate, and ethanesulfonate, arylsulfonates such as benzenesulfonate and p-toluenesulfonate, acetate, malate, fumarate, succinate, citrate, tartrate, oxalate, and maleate; and amino acid salts such as glycine salt, lysine salt, 25 arginine salt, ornithine salt, glutamate, and aspartate.

[0050] The "pharmacologically acceptable salt thereof" refers to a salt of the oligonucleotide analogue of the present invention, because the oligonucleotide can be

converted into the salt. Preferred examples of the salt are metal salts, for example, alkali metal salts such as sodium salt, potassium salt, and lithium salt, alkaline earth metal salts such as calcium salt and magnesium salt,

5 aluminum salt, iron salt, zinc salt, copper salt, nickel salt, and cobalt salt; amine salts, for example, inorganic salts such as ammonium salt, and organic salts such as t-octylamine salt, dibenzylamine salt, morpholine salt, glucosamine salt, phenylglycine alkyl ester salt,

10 ethylenediamine salt, N-methylglucamine salt, guanidine salt, diethylamine salt, triethylamine salt, dicyclohexylamine salt, N,N'-dibenzylethylenediamine salt, chloroprocaine salt, procaine salt, diethanolamine salt, N-benzyl-phenethylamine salt, piperazine salt,

15 tetramethylammonium salt, and tris(hydroxymethyl)aminomethane salt; inorganic acid salts, for example, halogenated hydroacid salts such as hydrofluoride, hydrochloride, hydrobromide, and hydriodide, nitrate, perchlorate, sulfate, and phosphate; organic acid

20 salts, for example, lower alkanesulfonates such as methanesulfonate, trifluoromethanesulfonate, and ethanesulfonate, arylsulfonates such as benzenesulfonate and p-toluenesulfonate, acetate, malate, fumarate, succinate, citrate, tartrate, oxalate, and maleate; and

25 amino acid salts such as glycine salt, lysine salt, arginine salt, ornithine salt, glutamate, and aspartate.

[0051] Of the compounds (I) of the present invention and the salts thereof, preferred compounds are as follows:

(1) Compounds and salts thereof, with R₁ being a hydrogen atom, an aliphatic acyl group, an aromatic acyl group, an aliphatic or aromatic sulfonyl group, a methyl group substituted by one to three aryl groups, a methyl group substituted by one to three aryl groups having an aryl ring substituted by a lower alkyl, lower alkoxy, halogen, or cyano group, or a silyl group.

(2) Compounds and salts thereof, with R₁ being a hydrogen atom, an acetyl group, a benzoyl group, a methanesulfonyl group, a p-toluenesulfonyl group, a benzyl group, a p-methoxybenzyl group, a trityl group, a dimethoxytrityl group, a monomethoxytrityl group, or a tert-butyldiphenylsilyl group.

(3) Compounds and salts thereof, with R₂ being a hydrogen atom, an aliphatic acyl group, an aromatic acyl group, an aliphatic or aromatic sulfonyl group, a methyl group substituted by one to three aryl groups, a methyl group substituted by one to three aryl groups having an aryl ring substituted by a lower alkyl, lower alkoxy, halogen, or cyano group, a silyl group, a phosphoroamidite group, a phosphoryl group, a phosphate group, or a phosphate group protected with a protective group for nucleic acid synthesis.

(4) Compounds and salts thereof, with R₂ being a hydrogen atom, an acetyl group, a benzoyl group, a methanesulfonyl group, a p-toluenesulfonyl group, a benzyl group, a p-methoxybenzyl group, a tert-butyldiphenylsilyl group, -P(OC₂H₄CN)(N(i-Pr)₂), -P(OCH₃)(N(i-Pr)₂), a

phosphonyl group, or a 2-chlorophenyl- or 4-chlorophenylphosphate group.

(5) Compounds and salts thereof, with R₃ being a hydrogen atom, an alkyl group having 1 to 5 carbon atoms, 5 an alkenyl group having 1 to 5 carbon atoms, an aryl group having 6 to 14 carbon atoms, a methyl group substituted by one to three aryl groups, a lower aliphatic or aromatic sulfonyl group such as a methanesulfonyl group or a p-toluenesulfonyl group, an aliphatic acyl group having 1 to 10 5 carbon atoms such as an acetyl group, or an aromatic acyl group such as a phenoxyacetyl group or a benzoyl group; or the compounds and salts thereof according to any one of claims 1 to 6, with the functional molecule unit substituent as R₃ being a fluorescent or chemiluminescent 15 labeling molecule, a nucleic acid incision activity functional group, or an intracellular or nuclear transfer signal peptide.

(6) Compounds and salts thereof, with Base being a 6-aminopurin-9-yl (i.e. adeninyl) group, a 6-aminopurin-9-20 yl group having the amino group protected with a protective group for nucleic acid synthesis, a 2,6-diaminopurin-9-yl group, a 2-amino-6-chloropurin-9-yl group, a 2-amino-6-chloropurin-9-yl group having the amino group protected with a protective group for nucleic acid synthesis, a 2-amino-6-fluoropurin-9-yl group, a 2-amino-6-fluoropurin-9-25 yl group having the amino group protected with a protective group for nucleic acid synthesis, a 2-amino-6-bromopurin-9-yl group, a 2-amino-6-bromopurin-9-yl group having the

amino group protected with a protective group for nucleic acid synthesis, a 2-amino-6-hydroxypurin-9-yl (i.e., guaninyl) group, a 2-amino-6-hydroxypurin-9-yl group having the amino group protected with a protective group for 5 nucleic acid synthesis, a 6-amino-2-methoxypurin-9-yl group, a 6-amino-2-chloropurin-9-yl group, a 6-amino-2-fluoropurin-9-yl group, a 2,6-dimethoxypurin-9-yl group, a 2,6-dichloropurin-9-yl group, a 6-mercaptopurin-9-yl group, a 2-oxo-4-amino-1,2-dihydropyrimidin-1-yl (i.e., cytosinyl) 10 group, a 2-oxo-4-amino-1,2-dihydropyrimidin-1-yl group having the amino group protected with a protective group for nucleic acid synthesis, a 2-oxo-4-amino-5-fluoro-1,2-dihydropyrimidin-1-yl group, a 2-oxo-4-amino-5-fluoro-1,2-dihydropyrimidin-1-yl group having the amino group 15 protected with a protective group for nucleic acid synthesis, a 4-amino-2-oxo-5-chloro-1,2-dihydropyrimidin-1-yl group, a 2-oxo-4-methoxy-1,2-dihydropyrimidin-1-yl group, a 2-oxo-4-mercpto-1,2-dihydropyrimidin-1-yl group, a 2-oxo-4-hydroxy-1,2-dihydropyrimidin-1-yl (i.e., uracinyl) 20 group, a 2-oxo-4-hydroxy-5-methyl-1,2-dihydropyrimidin-1-yl (i.e., thyminyl) group, a 4-amino-5-methyl-2-oxo-1,2-dihydropyrimidin-1-yl (i.e., 5-methylcytosinyl) group, or 4-amino-5-methyl-2-oxo-1,2-dihydropyrimidin-1-yl having the 25 amino group protected with a protective group for nucleic acid synthesis.

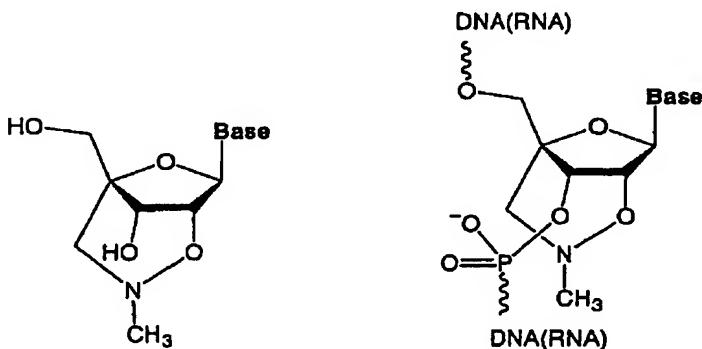
(7) There can also be named compounds and salts thereof, with Base being a benzoylaminopurin-9-yl, adenyl, 2-isobutyrylamino-6-hydroxypurin-9-yl, guaninyl, 2-oxo-4-

benzoylamino-1,2-dihydropyrimidin-1-yl, cytosinyl, 2-oxo-5-methyl-4-benzoylamino-1,2-dihydropyrimidin-1-yl, 5-methylcytosinyl, uracinyl, or thyminyl group.

[0052] The above (1) to (2), (3) to (4), or (6) to (7) represent more preferred compounds, as their numbers grow. Also preferred are compounds and salts thereof in which, in the general formula (1), R₁ is arbitrarily selected from (1) to (2), R₂ is arbitrarily selected from (3) to (4), R₃ is arbitrarily selected from (5), and Base is arbitrarily selected from (6) to (7); or compounds and salts thereof obtained by arbitrary combinations of these numbers. Particularly preferred combinations are (2)-(3)-(5)-(6), (2)-(3)-(5)-(7), (2)-(4)-(5)-(6), and (2)-(4)-(5)-(7).

[0053] There are named compounds of the general formula (I) and salts thereof, particularly preferably, compounds of the following formula and salts thereof.

[0054] [Chemical Formula 3]



2',4'-BNA^{NC} monomer An example of structure of 2',4'-BNA^{NC} modified portion in 2',4'-BNA^{NC} modified oligonucleotide

[0055] In the structural formula of the above group,

Base has the same meanings as offered above.

[0056] Of the oligonucleotide analogues containing one or two or more unit structures of the nucleoside analogue represented by the general formula (II) of the present 5 invention, and pharmacologically acceptable salts thereof, preferred examples are as follows:

(8) Oligonucleotide analogues, and pharmacologically acceptable salts thereof, with R₃ being a hydrogen atom, an alkyl group having 1 to 5 carbon atoms, an aralkyl group 10 such as a benzyl group, an aliphatic or aromatic acyl group such as an acetyl group or a benzoyl group, or an aliphatic or aromatic sulfonyl group such as a methanesulfonyl group or a p-toluenesulfonyl group,

(9) Oligonucleotide analogues, and pharmacologically acceptable salts thereof, with Base being a 6-aminopurin-9-yl (i.e. adeninyl) group, a 6-aminopurin-9-yl group having the amino group protected with a protective group for nucleic acid synthesis, a 2,6-diaminopurin-9-yl group, a 2-amino-6-chloropurin-9-yl group, a 2-amino-6-chloropurin-9-20 yl group having the amino group protected with a protective group for nucleic acid synthesis, a 2-amino-6-fluoropurin-9-yl group, a 2-amino-6-fluoropurin-9-yl group having the amino group protected with a protective group for nucleic acid synthesis, a 2-amino-6-bromopurin-9-yl group, a 2-amino-6-bromopurin-9-yl group having the amino group 25 protected with a protective group for nucleic acid synthesis, a 2-amino-6-hydroxypurin-9-yl (i.e., guaninyl) group, a 2-amino-6-hydroxypurin-9-yl group having the amino

group protected with a protective group for nucleic acid synthesis, a 6-amino-2-methoxypurin-9-yl group, a 6-amino-2-chloropurin-9-yl group, a 6-amino-2-fluoropurin-9-yl group, a 2,6-dimethoxypurin-9-yl group, a 2,6-5 dichloropurin-9-yl group, a 6-mercaptopurin-9-yl group, a 2-oxo-4-amino-1,2-dihydropyrimidin-1-yl (i.e., cytosinyl) group, a 2-oxo-4-amino-1,2-dihydropyrimidin-1-yl group having the amino group protected with a protective group for nucleic acid synthesis, a 2-oxo-4-amino-5-fluoro-1,2-10 dihydropyrimidin-1-yl group, a 2-oxo-4-amino-5-fluoro-1,2-dihydropyrimidin-1-yl group having the amino group protected with a protective group for nucleic acid synthesis, a 4-amino-2-oxo-5-chloro-1,2-dihydropyrimidin-1-yl group, a 2-oxo-4-methoxy-1,2-dihydropyrimidin-1-yl group, 15 a 2-oxo-4-mercpto-1,2-dihydropyrimidin-1-yl group, a 2-oxo-4-hydroxy-1,2-dihydropyrimidin-1-yl (i.e., uracinyl) group, a 2-oxo-4-hydroxy-5-methyl-1,2-dihydropyrimidin-1-yl (i.e., thyminyl) group, a 4-amino-5-methyl-2-oxo-1,2-dihydropyrimidin-1-yl (i.e., 5-methylcytosinyl) group, or 20 4-amino-5-methyl-2-oxo-1,2-dihydropyrimidin-1-yl having the amino group protected with a protective group for nucleic acid synthesis.

(10) There can also be named oligonucleotide analogues, and pharmacologically acceptable salts thereof, 25 with Base being a benzoylaminopurin-9-yl, adenyl, 2-isobutyrylamino-6-hydroxypurin-9-yl, guaninyl, 2-oxo-4-benzoylaminoo-1,2-dihydropyrimidin-1-yl, cytosinyl, 2-oxo-5-methyl-4-benzoylaminoo-1,2-dihydropyrimidin-1-yl,

5-methylcytosinyl, uracinyl, or thyminyl group.

[0057] The above (9) to (10) represent more preferred oligonucleotide analogues, as their numbers grow. Also preferred are oligonucleotide analogues and

5 pharmacologically acceptable salts thereof in which R₃ is arbitrarily selected from (8), Base is arbitrarily selected from (9) to (10), and the selected R₃ and Base are arbitrarily combined. Particularly preferred combinations in the general formula (II) are (8)-(9) and (8)-(10).

10 [0058] The nucleoside analogues and oligonucleotide analogues of the present invention can be synthesized based on the methods described in the Examples offered herein, and the conventional technologies in the field of the art.

(1) Synthesis of nucleoside analogues

15 Compounds represented by the general formula (I) can be synthesized based on the methods described in the Examples and the conventional technologies in the field of the art. The reaction conditions, reagents for introduction of a protective group, and reaction reagents
20 can be determined with reference to the methods described in the Examples. However, these methods are not limitative, and reaction conditions and reagents usable based on common knowledge in the field of the art can be employed as appropriate. For example, the methods described in
25 Japanese Patent Application Laid-Open No. 2000-297097 and Japanese Patent Application Laid-Open No. 1998-304889 can be referred to. Moreover, if the compounds of the general formula (I) or (II) have various natural or nonnatural

nucleic acid bases or other aromatic heterocyclic rings or aromatic hydrocarbon rings as Base, the starting materials for the compounds of the present invention can be synthesized with reference to the methods described in

5 Japanese Patent Application Laid-Open No. 1998-304889.

(2) Synthesis of oligonucleotide analogues

The oligonucleotide analogue containing the nucleoside analogue of the present invention can be synthesized in various forms by a publicly known DNA 10 synthesizer. Then, the resulting oligonucleotide analogue is purified by use of a reversed phase column, and the purity of the product is analyzed by reversed phase HPLC or MALDI-TOF-MS, whereby the formation of a purified oligonucleotide analogue can be confirmed.

15 One or more of the nucleoside analogues of the present invention can be rendered present in the oligonucleotide analogue. Furthermore, the nucleoside analogues may be rendered present at two or more locations of the oligonucleotide analogue in spaced state via one or 20 two or more natural nucleotides. According to the present invention, it is possible to synthesize an oligonucleotide analogue in which the nucleoside analogues of the present invention have been introduced in necessary numbers (length) at necessary positions. The length of the entire 25 oligonucleotide analogue is 2 to 50, preferably 8 to 30 nucleotide units.

[0059] The oligonucleotide analogue of the present invention is minimally degradable by nuclease and,

following administration in vivo, can exist in vivo for a long period. The oligonucleotide analogue forms a duplex with sense RNA, for example, to inhibit the transcription of an etiological in vivo component (protein) to mRNA. The 5 oligonucleotide analogue is also considered to inhibit the multiplication of an infecting virus.

[0060] In the light of these facts, the oligonucleotide analogues of the present invention can be expected to be useful as medicines, including antitumor agents and 10 antiviral agents, for inhibiting the actions of genes to treat diseases. That is, according to the present invention, there are provided oligonucleotide analogues, which have stable and excellent antisense or antigene activity or have excellent activities as reagents for 15 detection of particular genes, or as primers for initiation of amplification, or nucleoside analogues which are intermediates for production of the oligonucleotide analogues.

[0061] DNA or RNA oligonucleotide analogues 20 (oligonucleotide analogues), which have the 2',4'-BNA^{NC} monomer, as one of the nucleoside analogues of the present invention, modified in various forms, are useful as materials for various physiological and bioactive substances and pharmaceuticals, functional materials for 25 double-stranded oligonucleotides for the RNA interference method or decoy method, functional materials for DNA chips or molecular beacons targeting single-stranded nucleic acids such as cDNA, functional materials for applications

to various antisense methods (including ribozymes and DNAzymes), antigenic method, and genetic homologous recombination, materials for high sensitivity analysis of biological trace components by combinations with 5 fluorescent or luminescent substances, and starting materials for the development of research reagents for analysis and elucidation of gene functions.

[0062] The nucleoside analogues or oligonucleotide analogues of the present invention can be formed into 10 preparations for parenteral administration by incorporating therein customary adjuvants such as buffers and/or stabilizers. For use as preparations for topical administration, customary pharmaceutical carriers are incorporated, whereby the nucleoside analogues or 15 oligonucleotide analogues can be formed into ointments, creams, liquids and solution, or plasters.

[0063] The nucleoside analogues and oligonucleotide analogues of the present invention were synthesized in accordance with the synthesis scheme shown below. These 20 syntheses will be described in further detail in the Examples. Compounds 16, 21, 17 and 22 are amidites for introducing 2',4'-BNA^{NC}(NMe) and 2',4'-BNA^{NC}(NH), in which the nucleic acid base portions are thymine and methylcytosine, respectively, into oligonucleotides. From 25 compound 28 as well, amidite 29 for introduction of a methylcytidine BNA^{NC}(NH) derivative can be synthesized in the usual manner (tritylation, amiditation). Derivatives of Compound 1, in which the nucleic acid base portions are

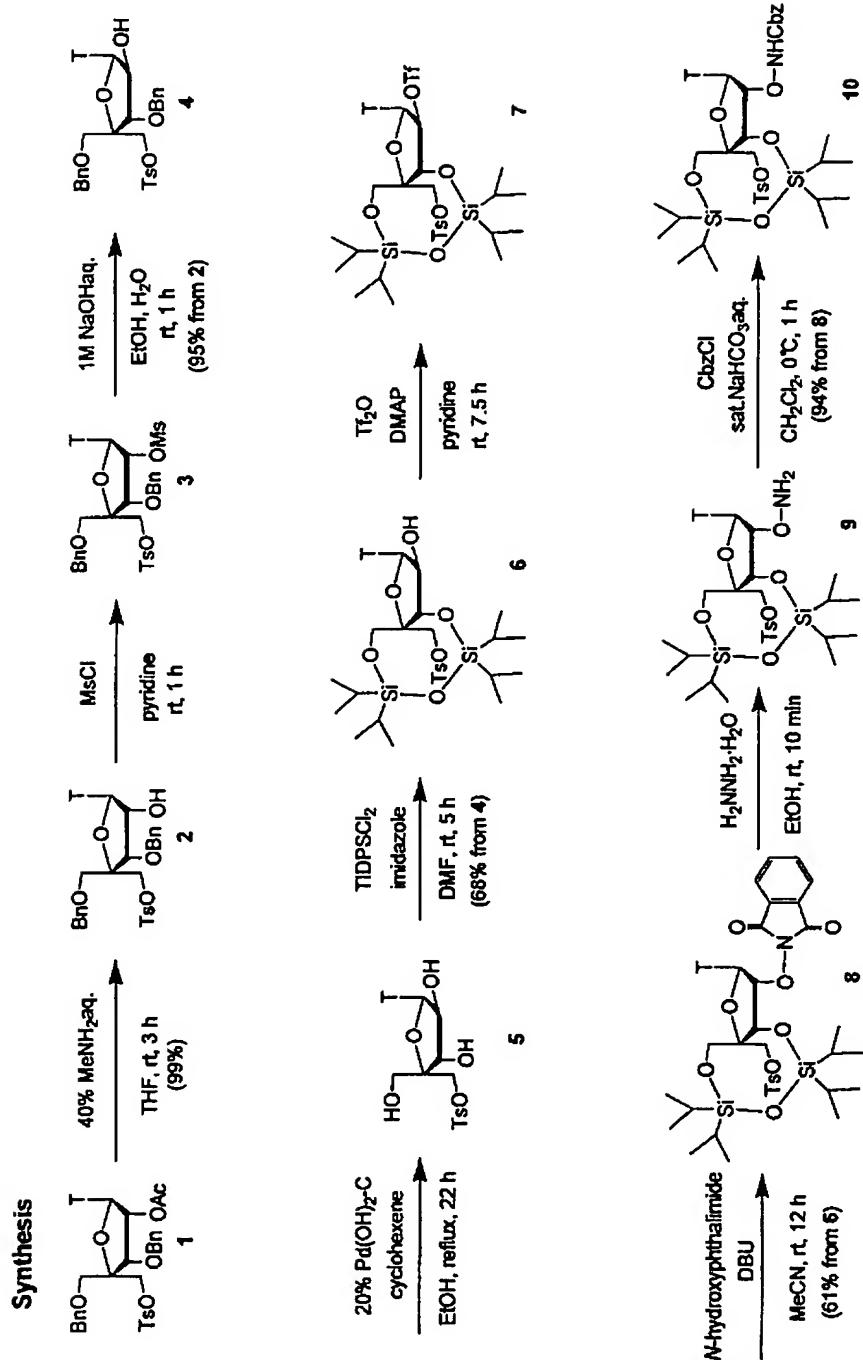
adenine and guanine instead of thymine, are also known. Thus, adenosine and guanosine 2',4'-BNA^{NC} derivatives are also considered to be synthesizable by this synthesis route.

Examples

5 [0064] The nucleoside analogues and oligonucleotide analogues of the present invention were synthesized in accordance with the synthesis scheme offered below. These syntheses will be described in further detail in the Examples. The characteristics of the synthesized
10 oligonucleotide analogues were measured by Experimental Examples.

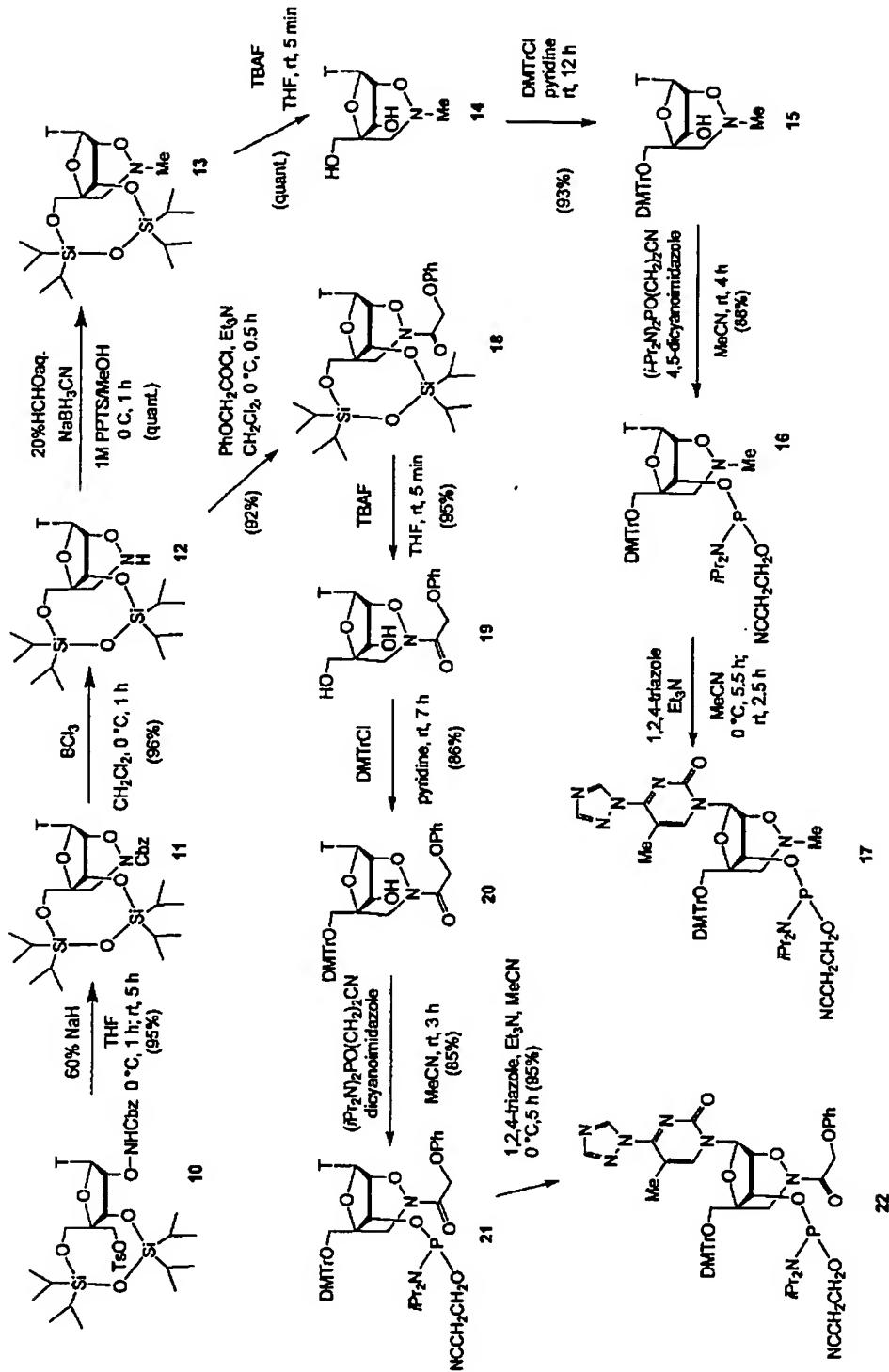
[0065]

[Chemical Formula 4]



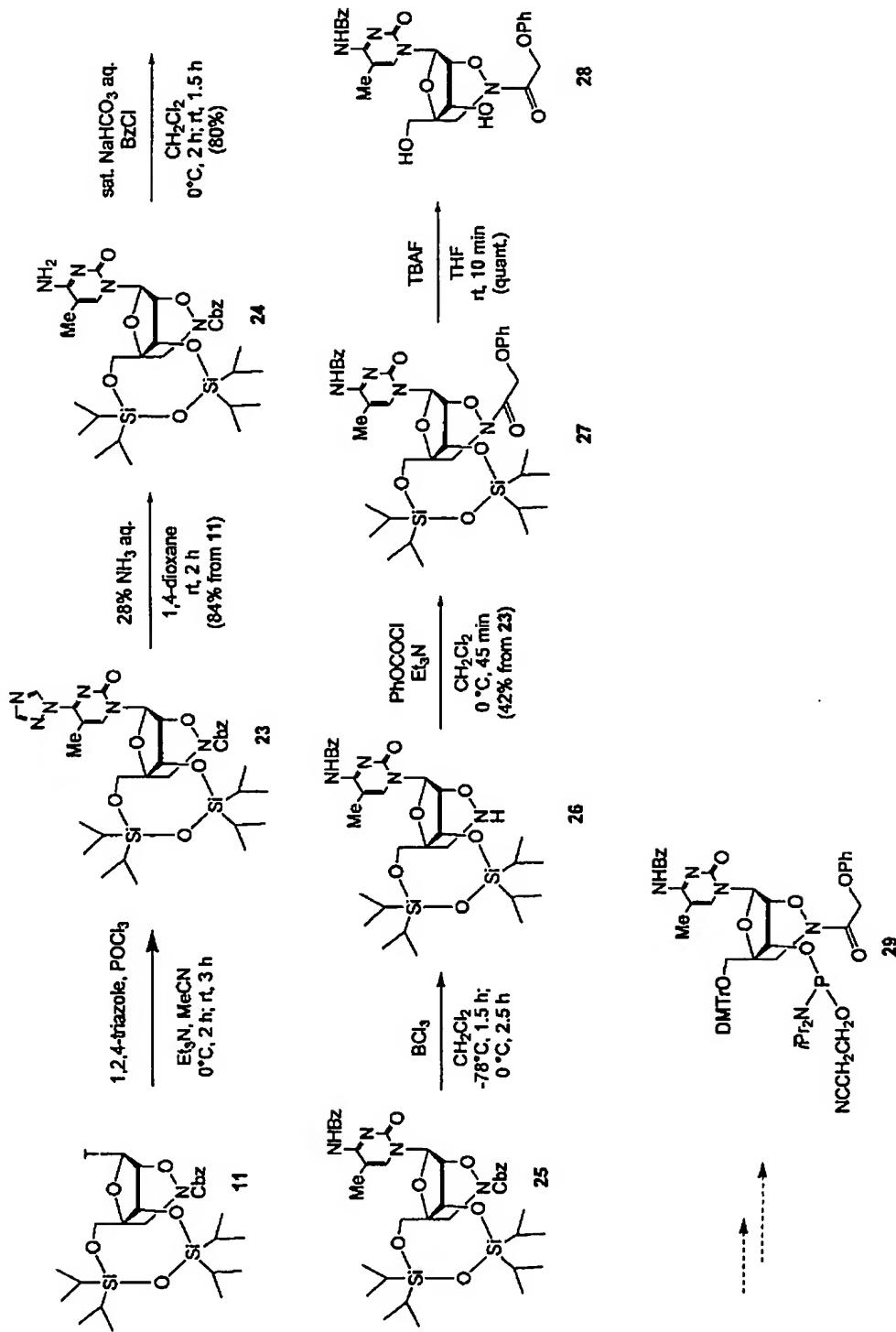
[0066]

[Chemical Formula 5]



[0067]

[Chemical Formula 6]



[0068]

[Example 1] Synthesis of nucleoside analogues (2',4'-BNA^{NC} amidites)

(1) Synthesis of Compound 2

5 A 40% aqueous solution of methylamine (0.11 ml, 1.50 mmols) was added to a tetrahydrofuran solution (3.5 ml) of Compound 1 (49 mg, 0.073 mmol) shown in the above synthesis scheme under ice-cooled conditions, and the mixture was stirred for 3 hours at room temperature. After the solvent 10 of the reaction solution was distilled off, the residue was extracted with ethyl acetate, and the organic layer was washed with water and a saturated aqueous solution of sodium chloride. The organic layer was dried over anhydrous sodium sulfate, and the solvent was distilled off.

15 Then, the residue was purified by silica gel column chromatography (n-hexane:ethyl acetate = 1:1) to obtain Compound 2 (45 mg, 99%) as a white solid.

[0069] [Equation 1]

mp 64-66°C. ¹H-NMR (CDCl₃) δ: 1.62 (3H, s), 2.43 (3H, s), 3.10 (1H, br), 3.63, 3.71 (2H, AB, J= 10 Hz), 4.18 (2H, s), 4.24 (1H, d, J= 6 Hz), 4.33 (1H, dd, J= 5, 6 Hz), 4.52 (1H, s), 4.60, 4.67 (2H, AB, J= 12 Hz), 5.68 (1H, d, J= 5 Hz), 7.23-7.37 (13H, m), 7.71 (2H, d, J= 8 Hz), 8.52 (1H, br).

20 [0070] (2) Synthesis of Compound 3

In a stream of nitrogen, methanesulfonyl chloride (45 ml, 0.59 mmol) was added to a pyridine solution (1.5 ml) of Compound 2 (146 mg, 0.23 mmol) under ice-cooled conditions, and the mixture was stirred for 1 hour at room temperature. 25 Water was added to the reaction solution, and the mixture was extracted with ethyl acetate. The organic layer was

washed with a saturated aqueous solution of sodium bicarbonate, and a saturated aqueous solution of sodium chloride. Then, the organic layer was dried over anhydrous sodium sulfate. The solvent was distilled off under reduced pressure, and the resulting crude product 3 (170 mg) was used for a next reaction without being purified.

[0071] [Equation 2]

$^1\text{H-NMR}$ (CDCl_3) δ : 1.65 (3H, s), 2.43 (3H, s), 3.07 (3H, s), 3.55, 3.81 (2H, AB, J = 10 Hz), 4.13, 4.22 (2H, AB, J = 11 Hz), 4.41 (1H, d, J = 6 Hz), 4.48 (2H, s), 4.48, 4.78 (2H, AB, J = 12 Hz), 5.32 (1H, dd, J = 4, 6 Hz), 5.91 (1H, d, J = 4Hz), 7.20-7.38 (13H, m), 7.72 (2H, d, J = 6 Hz).

[0072] (3) Synthesis of Compound 4

A 1M aqueous solution of sodium hydroxide (0.70 ml, 0.70 mmol) was added, at room temperature, to a water-ethanol solution (1:2, 6 ml) of the crude product 3 (170 mg) obtained in the preceding reaction, and the mixture was stirred for 1 hour. After neutralization with a 10% aqueous solution of hydrochloric acid, the mixture was extracted with ethyl acetate. The organic layer was washed with water and a saturated aqueous solution of sodium chloride, and then dried over anhydrous sodium sulfate. The solvent was distilled off under reduced pressure, and the resulting crude product was purified by silica gel column chromatography (chloroform:methanol = 15:1) to obtain Compound 4 (139 mg, 95% from 2 in 2 steps) as a white solid.

[0073] [Equation 3]

mp 73-76°C. $^1\text{H-NMR}$ (CDCl_3) δ : 1.65 (3H, s), 2.38 (3H, s), 3.64, 3.70 (2H, AB, $J= 9$ Hz), 4.12 (1H, d, $J= 2$ Hz), 4.15, 4.21 (2H, AB, $J= 10$ Hz), 4.43, 4.64 (2H, AB, $J= 11$ Hz), 4.49 (2H, s), 4.55 (1H, dd, $J= 2, 4$ Hz), 5.97 (1H, d, $J= 4$ Hz), 7.19-7.36 (13H, m), 7.73 (2H, d, $J= 8$ Hz), 9.72 (1H, br).

[0074] (4) Synthesis of Compound 5

In a stream of nitrogen, 20% palladium hydroxide-carbon powder (0.60 g) and cyclohexene (5.2 ml, 51 mmols) were added to an ethanol solution (10 ml) of Compound 4 (0.80 g, 1.28 mmols), and the mixture was refluxed with heating for 5 hours. Further, palladium hydroxide-carbon powder (0.20 g) was added, and the mixture was refluxed with heating for 17 hours. After the reaction solution was filtered, the solvent was distilled off under reduced pressure. The resulting crude product 5 (0.46 g) was used for a next reaction without being purified.

[0075] [Equation 4]

mp 103-106°C. $^1\text{H-NMR}$ (CDCl_3) δ : 1.85 (3H, d, $J= 1$ Hz), 2.41 (3H, s), 3.67 (2H, s), 4.14, 4.25 (2H, AB, $J= 11$ Hz), 4.16 (1H, dd, $J= 4, 5$ Hz), 4.21 (1H, d, $J= 4$ Hz), 5.87 (1H, d, $J= 5$ Hz), 7.38 (2H, d, $J= 8$ Hz), 7.48 (1H, d, $J= 1$ Hz), 7.79 (2H, d, $J= 8$ Hz).

[0076] (5) Synthesis of Compound 6

In a stream of nitrogen, 1,3-dichloro-1,1,3,3-tetraisopropylsiloxyane (0.45 ml, 1.41 mmols) and imidazole (0.38 g, 5.63 mmols) were added to an N,N-dimethylformamide solution (10 ml) of Compound 5 (0.46 g), and the mixture was stirred for 5 hours at room temperature. The reaction mixture was extracted with ether, and the organic layer was washed with water and a saturated aqueous solution of sodium chloride, followed by drying the organic layer over magnesium sulfate. The solvent was distilled off under reduced pressure, and the resulting crude product

was purified by silica gel column chromatography (n-hexane:ethyl acetate = 2:1→1:1) to obtain Compound 6 (0.60 g, 68% from 4 in 2 steps) as a white solid.

[0077] [Equation 5]

mp 97-99°C. IR ν_{max} (KBr): 3185, 3058, 2947, 2869, 1697, 1467, 1366, 1277, 1182, 1036, 1081, 1036 cm⁻¹. ¹H-NMR (CDCl₃) δ: 0.82-1.29 (28H, m), 1.86 (3H, s), 2.45 (3H, s), 3.73, 4.01 (2H, AB, J = 12 Hz), 4.06, 4.29 (2H, AB, J = 11 Hz), 4.47 (1H, d, J = 7 Hz), 4.52 (1H, dd, J = 6, 7 Hz), 5.98 (1H, d, J = 6 Hz), 7.28 (1H, s), 7.43 (2H, d, J = 8 Hz), 7.81 (2H, d, J = 8 Hz).

5

[0078] (6) Synthesis of Compound 7

In a stream of nitrogen, trifluoromethanesulfonic anhydride (0.15 ml, 0.88 mmol) and 4-(dimethylamino)pyridine (7 mg, 0.06 mmol) were added to a pyridine solution (3 ml) of Compound 6 (200 mg, 0.29 mmol) under ice-cooled conditions, and the mixture was stirred for 7.5 hours at room temperature. Water was added to the reaction solution, and the mixture was extracted with dichloromethane. The organic layer was washed with a saturated aqueous solution of sodium bicarbonate, and a saturated aqueous solution of sodium chloride. Then, the organic layer was dried over anhydrous sodium sulfate. The solvent was distilled off under reduced pressure, and the resulting crude product 7 (0.29 g) was used for a next reaction without being purified.

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[0079] [Equation 6]

IR ν_{max} (KBr): 3178, 3069, 2948, 2874, 1695, 1465, 1424, 1365, 1281, 1214, 1143, 1088, 1038 cm⁻¹. ¹H-NMR (CDCl₃) δ: 0.81-1.26 (28H, m), 1.92 (3H, s), 2.46 (3H, s), 3.77, 4.16 (2H, AB, J = 12 Hz), 4.15, 4.42 (2H, AB, J = 10 Hz), 5.42 (1H, d, J = 7 Hz), 5.45 (1H, d, J = 7 Hz), 6.89 (1H, s), 7.37 (2H, d, J = 8 Hz), 7.82 (2H, d, J = 8 Hz), 7.92 (1H, s).

[0080] (7) Synthesis of Compound 8

In a stream of nitrogen, N-hydroxyphthalimide (67 mg, 0.41 mmol) and 1,8-diazabicyclo[5.4.0]-7-undecene (61 ml, 0.41 mmol) were added, at room temperature, to an acetonitrile solution (3 ml) of Crude Product 7 (0.29 g) obtained in the preceding reaction, and the mixture was stirred for 12 hours at room temperature. The reaction solution was extracted with dichloromethane, and the organic layer was washed with water and a saturated aqueous solution of sodium chloride, followed by drying the organic layer over anhydrous sodium sulfate. The solvent was distilled off under reduced pressure, and the resulting crude product was purified by silica gel column chromatography (chloroform) to obtain Compound 8 (0.15 g, 61% from 6 in 2 steps) as a yellow solid.

15 [0081] [Equation 7]

mp 109-111°C. $^1\text{H-NMR}$ (CDCl_3) δ : 0.99-1.31 (28H, m), 1.87 (3H, s), 1.42 (3H, s), 3.88, 4.19 (2H, AB, $J=13$ Hz), 4.41, 4.91 (2H, AB, $J=12$ Hz), 4.77 (1H, d, $J=6$ Hz), 4.80 (1H, d, $J=6$ Hz), 6.00 (1H, s), 7.32-7.37 (3H, m), 7.74-7.90 (6H, m).

[0082] (8) Synthesis of Compound 9

Hydrazine hydrate (0.12 ml, 2.38 mmols) was added to an ethanol solution (35 ml) of Compound 8 (1.16 g, 1.40 mmols), and the mixture was stirred for 10 minutes at room temperature. After the solvent of the reaction solution was distilled off, the residue was filtered, and the filtrate was extracted with ethyl acetate. The organic layer was washed with water and a saturated aqueous solution of sodium chloride. Then, the organic layer was dried over anhydrous sodium sulfate. The solvent was distilled off under reduced pressure, and the resulting

crude product 9 (0.93 g) was used for a next reaction without being purified.

[0083] [Equation 8]

mp 117-119°C. $^1\text{H-NMR}$ (CDCl_3) δ : 0.87-1.09 (28H, m), 1.92 (3H, d, J = 1 Hz), 2.45 (3H, s), 3.83, 3.91 (2H, AB, J = 12 Hz), 4.22, 4.57 (2H, AB, J = 16 Hz), 4.41 (1H, dd, J = 1, 7 Hz), 4.80 (1H, d, J = 7 Hz), 5.35 (1H, d, J = 1 Hz), 5.81 (1H, br), 7.12 (1H, d, J = 1 Hz), 7.33 (2H, d, J = 8 Hz), 7.81 (2H, d, J = 8 Hz), 8.75 (1H, s).

5 [0084] (9) Synthesis of Compound 10

In a stream of nitrogen, a saturated aqueous solution of sodium bicarbonate (4.0 ml, 4.2 mmols) and benzyl chloroformate (0.30 ml, 2.1 mmols) were added, under ice-cooled conditions, to a methylene chloride solution (15 ml) of Crude Product 9 (0.93 g) obtained in the preceding reaction, and the mixture was stirred for 1 hour. A saturated aqueous solution of sodium bicarbonate was added to the reaction solution, and the mixture was extracted with ethyl acetate. The organic layer was washed with water and a saturated aqueous solution of sodium chloride, followed by drying the organic layer over magnesium sulfate. The solvent was distilled off under reduced pressure, and the resulting crude product was purified by silica gel column chromatography (n-hexane:ethyl acetate = 4:1) to obtain Compound 10 (0.92 g, 94% from 8 in 2 steps) as a white solid.

[0085] [Equation 9]

mp 82-84°C. IR ν_{max} (KBr): 3319, 3059, 2948, 2874, 1691, 1464, 1365, 1245, 1180, 1098, 1034 cm^{-1} . $^1\text{H-NMR}$ (CDCl_3) δ : 0.86 (28H, m), 1.87 (3H, s), 2.43 (3H, s), 3.80, 3.88 (2H, AB, J = 11 Hz), 4.25, 4.53 (2H, AB, J = 11 Hz), 4.72 (1H, d, J = 6 Hz), 5.02 (1H, d, J = 6 Hz), 5.15, 5.20 (2H, AB, J = 12 Hz), 5.37 (1H, s), 7.31 (2H, d, J = 8 Hz), 7.34 (5H, s), 7.80 (2H, d, J = 8 Hz), 7.95 (1H, s), 8.48 (1H, s).

[0086] (10) Synthesis of Compound 11

In a stream of nitrogen, a tetrahydrofuran solution (15 ml) Compound 10 (3.81 g, 4.57 mmols) was added dropwise, under ice-cooled conditions, to a tetrahydrofuran suspension (25 ml) of sodium hydride (60% in oil, 0.55 g, 13.7 mmols), and the mixture was stirred for 1 hour, followed by stirring the mixture for 5 hours at room temperature. After neutralization in a saturated aqueous solution of oxalic acid, the reaction mixture was extracted with ethyl acetate. The organic layer was washed with water and a saturated aqueous solution of sodium chloride, followed by drying the organic layer over anhydrous sodium sulfate. The solvent was distilled off under reduced pressure, and the resulting crude product was purified by silica gel column chromatography (chloroform → chloroform → methanol = 100:1) to obtain Compound 11 (2.87 g, 95%) as a white solid.

[0087] [Equation 10]

mp 61-64°C. IR ν_{max} (KBr): 3178, 3026, 2941, 2864, 1697, 1463, 1271, 1221, 1164, 1073 cm⁻¹. ¹H-NMR (CDCl₃) δ: 0.80-1.12 (28H, m), 1.92 (3H, s), 3.62, 3.85 (2H, AB, *J* = 12 Hz), 3.68, 4.11 (2H, AB, *J* = 13 Hz), 4.11 (1H, d, *J* = 4 Hz), 4.55 (1H, d, *J* = 4 Hz), 5.21, 5.28 (2H, AB, *J* = 12 Hz), 5.98 (1H, s), 7.30-7.44 (5H, m), 7.61 (1H, s), 8.80 (1H, br).

[0088] (11) Synthesis of Compound 12

In a stream of nitrogen, a 1M hexane solution of boron trichloride (5.29 ml, 5.29 mmols) was added, under ice-cooled conditions, to a methylene chloride solution (10 ml) of Crude Product 11 (0.35 mg, 0.53 mmol) obtained in the preceding reaction, and the mixture was stirred for 1 hour. A saturated aqueous solution of sodium bicarbonate

was added to the reaction solution, and the mixture was extracted with ethyl acetate. The organic layer was washed with water and a saturated aqueous solution of sodium chloride, followed by drying the organic layer over 5 anhydrous sodium sulfate. The solvent was distilled off under reduced pressure, and the resulting crude product was purified by silica gel column chromatography (chloroform:methanol = 50:1) to obtain Compound 12 (0.27 g, 96%) as a white solid.

10 [0089] [Equation 11]

mp 102-104°C. IR ν_{max} (KBr): 4326, 3178, 3059, 2947, 2864, 1698, 1564, 1464, 1270, 1043 cm^{-1} . $^1\text{H-NMR}$ (CDCl_3) δ : 0.99-1.25 (2H, m), 1.93 (3H, s), 2.54, 3.68 (2H, AB, J = 13 Hz), 3.67, 4.06 (2H, AB, J = 13 Hz), 4.10 (1H, d, J = 2 Hz), 4.36 (1H, d, J = 2 Hz), 6.17 (1H, s), 7.73 (1H, s), 8.71 (1H, br).

[0090] (12) Synthesis of Compound 13

A 20% aqueous solution of formaldehyde (0.06 ml, 0.40 mmol) was added to a 1M pyridinium p-toluenesulfonate-15 methanol solution (3.6 ml) of Compound 12 (0.19 g, 0.36 mmol) at room temperature, and the mixture was stirred for 10 minutes. Further, sodium cyanoborohydride (45 mg, 0.72 mmol) was added under ice-cooled conditions, and the mixture was stirred for 1 hour. The reaction solution was 20 extracted with ethyl acetate, the extract was washed with water, a saturated aqueous solution of sodium bicarbonate, and a saturated aqueous solution of sodium chloride, and the organic layer was dried over anhydrous sodium sulfate. The solvent was distilled off under reduced pressure, and 25 the resulting crude product was purified by silica gel column chromatography (n-hexane:ethyl acetate = 2:1) to

obtain Compound 13 (0.19 g, 100%) as a white solid.

[0091] [Equation 12]

IR ν_{max} (KBr): 2947, 2868, 1695, 1464, 1267, 1163, 1038 cm⁻¹. ¹H-NMR (CDCl₃) δ : 0.94-1.12 (28H, m), 1.92 (3H, d, J = 1 Hz), 2.60, 2.91 (2H, AB, J = 11 Hz), 2.76 (3H, s), 3.67, 4.08 (2H, AB, J = 13 Hz), 3.95 (1H, d, J = 3 Hz), 4.33 (1H, d, J = 3 Hz), 6.24 (1H, s), 7.73 (1H, d, J = 1 Hz), 8.56 (1H, br).

[0092] (13) Synthesis of Compound 14

5 Tetra-n-butylammonium fluoride (1M in tetrahydrofuran, 0.17 ml, 0.17 mmol) was added to a tetrahydrofuran solution (2 ml) of Compound 13 (46 mg, 0.085 mmol), and the mixture was stirred for 5 minutes at room temperature. The solvent was distilled off under reduced pressure, and the resulting 10 crude product was purified by silica gel column chromatography (ethyl acetate:methanol = 15:1) to obtain Compound 14 (25 mg, 100%) as a white solid.

[0093] [Equation 13]

mp 101-103°C. ¹H-NMR (CD₃OD) δ : 1.85 (3H, d, J = 1 Hz), 2.71 (3H, s), 2.71, 2.88 (2H, AB, J = 11 Hz), 3.69, 3.76 (2H, AB, J = 12 Hz), 3.94 (1H, d, J = 3 Hz), 4.23 (1H, d, J = 3 Hz), 6.21 (1H, s), 7.99 (1H, d, J = 1 Hz).

15 [0094] (14) Synthesis of Compound 15

To a pyridine solution (10 ml) of Compound 14 (0.16 g, 0.54 mmol), 4,4'-dimethoxytrityl chloride (0.22 g, 0.64 mmol) was added, and the mixture was stirred for 12 hours at room temperature. A saturated aqueous solution of 20 sodium bicarbonate was added to the reaction mixture, and the mixture was extracted with ethyl acetate. The organic layer was washed with water and a saturated aqueous solution of sodium chloride, followed by drying the organic layer over anhydrous sodium sulfate. The solvent was 25 distilled off under reduced pressure, and the resulting

crude product was purified by silica gel column chromatography (1% triethylamine-containing n-hexane:ethyl acetate = 1:2 → ethyl acetate:methanol = 30:1) to obtain Compound 15 (0.30 g, 93%) as a white solid.

5 [0095] [Equation 14]

mp 133-134°C. $^1\text{H-NMR}$ (CDCl_3) δ : 1.45 (3H, d, J = 1 Hz), 2.52 (1H, d, J = 9 Hz), 2.72 (3H, s), 2.77 (2H, s), 3.32, 3.37 (2H, AB, J = 11 Hz), 3.79 (6H, s), 4.23 (1H, dd, J = 3, 9 Hz), 4.35 (1H, d, J = 3 Hz), 6.35 (1H, s), 6.84 (4H, d, J = 8 Hz), 7.22-7.46 (9H, m), 7.75 (1H, d, J = 1 Hz), 8.25 (1H, br).

[0096] (15) Synthesis of Compound 16

To an acetonitrile solution (6 ml) of Compound 15 (0.17 g, 0.28 mmol) and 4,5-dicyanoimidazole (40 mg, 10 0.34 mmol), 2-cyanoethyl-N,N,N',N'-tetraisopropylphosphoroamidite (0.13 ml, 0.42 mmol) was added, and the mixture was stirred for 4 hours at room temperature. A saturated aqueous solution of sodium bicarbonate was added to the reaction mixture, and the 15 mixture was extracted with ethyl acetate. The organic layer was washed with a saturated aqueous solution of sodium bicarbonate, water, and a saturated aqueous solution of sodium chloride, followed by drying the organic layer over anhydrous sodium sulfate. The solvent was distilled 20 off under reduced pressure, and the resulting crude product was purified by silica gel column chromatography (1% triethylamine-containing n-hexane:ethyl acetate = 1:1), followed by reprecipitation (ethyl acetate-hexane) to obtain Compound 16 (0.20 g, 88%) as a white solid.

25 [0097] [Equation 15]

mp 65-65°C. $^{31}\text{P-NMR}$ (acetone-d_6) δ : 148.69, 149.82.

[0098] (16) Synthesis of Compound 17

In a stream of nitrogen, phosphoryl chloride (86 ml, 0.92 mmol) was added, under ice-cooled conditions, to an acetonitrile suspension (9 ml) of 1,2,4-triazole (278 mg, 5 4.03 mmols), and the mixture was stirred vigorously for 10 minutes. Further, triethylamine (0.64 ml, 4.62 mmols) was added, and the mixture was stirred for 35 minutes. Under ice-cooled conditions, an acetonitrile solution (3 ml) of Compound 16 (95 mg, 0.12 mmol) was added, and the mixture 10 was stirred for 5.5 hours. Then, the mixture was further stirred for 2.5 hours at room temperature. A saturated aqueous solution of sodium bicarbonate was added to the reaction mixture, and the mixture was extracted with ethyl acetate. The organic layer was washed with water and a 15 saturated aqueous solution of sodium chloride, followed by drying the organic layer over anhydrous sodium sulfate. The solvent was distilled off under reduced pressure, and the resulting crude product was purified by silica gel column chromatography (n-hexane:ethyl acetate = 1:2), and 20 then reprecipitation (ethyl acetate-hexane) to obtain Compound 17 (83 mg, 83%) as a white solid.

[0099] [Equation 16]

mp 107-109°C. ^{31}P -NMR (acetone-d₆) δ: 148.24, 149.91.

[0100] (17) Synthesis of Compound 18

25 Phenoxyacetyl chloride (29 ml, 0.21 mmol) was added, under ice-cooled conditions, to a methylene chloride solution (3 ml) of Compound 12 (100 mg, 0.19 mmol) and triethylamine (32 ml, 0.12 mmol), and the mixture was

stirred for 30 minutes. A saturated aqueous solution of sodium bicarbonate was added to the reaction mixture, and the mixture was extracted with ethyl acetate. The organic layer was washed with water and a saturated aqueous 5 solution of sodium chloride, followed by drying the organic layer over anhydrous sodium sulfate. The solvent was distilled off under reduced pressure, and the resulting crude product was purified by silica gel column chromatography (n-hexane:ethyl acetate = 2:1) to obtain 10 Compound 18 (115 mg, 92%) as a white solid.

[0101] [Equation 17]

mp 124-126°C. $^1\text{H-NMR}$ (CDCl_3) δ : 1.02-1.13 (28H, m), 1.92 (3H, s), 3.49 (1H, AB, J = 13 Hz), 3.73 (1H, AB, J = 14 Hz), 4.16 (1H, d, J = 3 Hz), 4.18 (1H, AB, J = 13 Hz), 4.23 (1H, AB, J = 14 Hz), 4.57 (1H, d, J = 3 Hz), 4.91 (1H, AB, J = 16 Hz), 5.00 (1H, AB, J = 16 Hz), 5.97 (1H, s), 6.92-7.00 (3H, m), 7.24-7.30 (2H, m), 7.60 (1H, s), 9.59 (1H, brs).

[0102] (18) Synthesis of Compound 19

In a stream of nitrogen, tetra-n-butylammonium 15 fluoride (1M tetrahydrofuran solution, 1.0 ml, 1.0 mmol) was added to a tetrahydrofuran solution (10 ml) of Compound 18 (0.34 g, 0.51 mmol) at room temperature, and the mixture was stirred for 5 minutes. The solvent was distilled off under reduced pressure, and the resulting crude product was 20 purified by silica gel column chromatography (ethyl acetate:methanol = 15:1) to obtain Compound 19 (0.20 g, 95%) as a white solid.

[0103] [Equation 18]

mp 144-146°C. $^1\text{H-NMR}$ (CD_3OD) δ : 1.85 (3H, s), 3.53 (1H, AB, J = 13 Hz), 3.77 (1H, AB, J = 13 Hz), 3.85 (1H, AB, J = 13 Hz), 4.18 (1H, AB, J = 13 Hz), 4.20 (1H, d, J = 2 Hz), 4.60 (1H, d, J = 2 Hz), 4.93 (1H, AB, J = 16 Hz), 5.04 (1H, AB, J = 16 Hz), 6.05 (1H, s), 6.91-6.97 (3H, m), 7.23-7.28 (2H, m), 7.90 (1H, s).

[0104] (19) Synthesis of Compound 20

In a stream of nitrogen, 4,4'-dimethoxytrityl chloride (0.25 g, 0.73 mmol) was added to a pyridine solution (8 ml) of Compound 19 (0.17 g, 0.41 mmol) at room temperature, and the mixture was stirred for 7 hours. A saturated aqueous solution of sodium bicarbonate was added to the reaction mixture, and the mixture was extracted with ethyl acetate. The organic layer was washed with water and a saturated aqueous solution of sodium chloride, followed by drying the organic layer over anhydrous sodium sulfate. The solvent was distilled off under reduced pressure, and the resulting crude product was purified by silica gel column chromatography (1% triethylamine-containing n-hexane:ethyl acetate = 1:1 → ethyl acetate:methanol = 50:1) to obtain Compound 20 (0.25 g, 86%) as a white solid.

[0105] [Equation 19]

mp 144-145°C. $^1\text{H-NMR}$ (CDCl_3) δ : 1.44 (3H, s), 2.65 (1H, br), 3.40 (1H, AB, J = 11 Hz), 3.44 (1H, AB, J = 13 Hz), 3.52 (1H, AB, J = 11 Hz), 3.80 (6H, s), 4.23 (1H, AB, J = 13 Hz), 4.44 (1H, d, J = 3 Hz), 4.63 (1H, d, J = 3 Hz), 4.86 (1H, AB, J = 16 Hz), 4.99 (1H, AB, J = 16 Hz), 6.04 (1H, s), 6.83-6.98 (7H, m), 7.25-7.43 (11H, m), 7.65 (1H, s), 8.38 (1H, br).

[0106] (20) Synthesis of Compound 21

To an acetonitrile solution (7 ml) of Compound 20 (0.25 g, 0.35 mmol) and 4,5-dicyanoimidazole (41 mg, 0.35 mmol), 2-cyanoethyl-N,N,N',N'-tetraisopropylphosphoroamidite (0.13 ml, 0.42 mmol) was added, and the mixture was stirred for 3 hours at room temperature. A saturated aqueous solution of sodium bicarbonate was added to the reaction mixture, and the mixture was extracted with ethyl acetate. The organic

layer was washed with a saturated aqueous solution of sodium bicarbonate, water, and a saturated aqueous solution of sodium chloride, followed by drying the organic layer over anhydrous sodium sulfate. The solvent was distilled off under reduced pressure, and the resulting crude product was purified by silica gel column chromatography (1% triethylamine-containing n-hexane:ethyl acetate = 1:1), followed by reprecipitation (ethyl acetate-hexane) to obtain Compound 21 (0.27 g, 85%) as a white solid.

10 [0107] [Equation 20]

mp 107-109°C. ^{31}P -NMR (acetone- d_6) δ : 149.79, 151.04.

[0108] (21) Synthesis of Compound 22

In a stream of nitrogen, phosphoryl chloride (71 ml, 0.76 mmol) was added, under ice-cooled conditions, to an acetonitrile suspension (10 ml) of 1,2,4-triazole (229 mg, 3.32 mmols), and the mixture was stirred vigorously for 10 minutes. Further, triethylamine (0.53 ml, 3.81 mmols) was added, and the mixture was stirred for 35 minutes. Under ice-cooled conditions, an acetonitrile solution (2 ml) of Compound 21 (90 mg, 0.10 mmol) was added, and the mixture was stirred for 5 hours. A saturated aqueous solution of sodium bicarbonate was added to the reaction mixture, and the mixture was extracted with ethyl acetate. The organic layer was washed with water and a saturated aqueous solution of sodium chloride, followed by drying the organic layer over anhydrous sodium sulfate. The solvent was distilled off under reduced pressure, and the resulting crude product was purified by reprecipitation (ethyl

acetate-hexane) to obtain Compound 22 (90 mg, 95%) as a white solid.

[0109] [Equation 21]

mp 104-106°C. ^{31}P -NMR (acetone-*d*₆) δ : 149.58, 151.29.

5 [0110] (22) Synthesis of Compound 23

In a stream of nitrogen, phosphoryl chloride (0.44 ml, 4.67 mmols) was added, under ice-cooled conditions, to an acetonitrile suspension (61 ml) of 1,2,4-triazole (1.41 g, 20.4 mmols), and the mixture was stirred vigorously for 10 minutes. Further, triethylamine (3.27 ml, 23.4 mmols) was added, and the mixture was stirred for 35 minutes. Under ice-cooled conditions, an acetonitrile solution (3 ml) of Compound 11 (409 mg, 0.62 mmol) was added, and the mixture was stirred for 2 hours, and further stirred for 3 hours at room temperature. A saturated aqueous solution of sodium bicarbonate was added to the reaction mixture, and the mixture was extracted with ethyl acetate. The organic layer was washed with water and a saturated aqueous solution of sodium chloride, followed by drying the organic layer over anhydrous sodium sulfate. The solvent was distilled off under reduced pressure, and the resulting crude product 23 (497 mg) was used for a next reaction without being purified.

[0111] [Equation 22]

$^1\text{H-NMR}$ (CDCl_3) δ : 0.92-1.12 (28H, m), 2.47 (3H, s), 3.66 (1H, AB, J = 12 Hz), 3.72 (1H, AB, J = 13 Hz), 3.90 (1H, AB, J = 12 Hz), 4.06 (1H, d, J = 3 Hz), 4.16 (1H, AB, J = 13 Hz), 4.73 (1H, d, J = 3 Hz), 6.11 (1H, s), 7.25-7.48 (5H, m), 8.12 (1H, s), 8.24 (1H, s), 9.32 (1H, s).

25 [0112] (23) Synthesis of Compound 24

To a 1,4-dioxane solution (10.6 ml) of Compound 23 (497 mg), a 28% aqueous solution of ammonia (1.76 ml) was added, and the mixture was stirred for 2 hours at room temperature. The solvent was distilled off under reduced pressure, and the resulting crude product was purified by silica gel column chromatography (chloroform:methanol = 5 30:1) to obtain Compound 24 (343 mg, 84% from 11 in 2 steps) as a white solid.

[0113] [Equation 23]

mp 137-139°C. $^1\text{H-NMR}$ (CDCl_3) δ : 0.93-1.11 (28H, m), 2.03 (3H, s), 3.62 (1H, AB, J = 12 Hz), 3.69 (1H, AB, J = 13 Hz), 3.85 (1H, AB, J = 12 Hz), 4.03 (1H, d, J = 3 Hz), 4.11 (1H, AB, J = 13 Hz), 4.59 (1H, d, J = 3 Hz), 5.24 (2H, s), 6.03 (1H, s), 7.26-7.45 (5H, m), 7.74 (1H, s).

[0114] (24) Synthesis of Compound 25

A saturated aqueous solution of sodium bicarbonate (0.8 ml, 0.79 mmol) and benzoyl chloride (92 ml, 0.79 mmol) were added to a methylene chloride solution (2.6 ml) of 10 Compound 24 (175 mg, 0.26 mmol) under ice-cooled conditions, and the mixture was stirred for 2 hours, and further stirred for 1.5 hours at room temperature. A saturated aqueous solution of sodium bicarbonate was added to the reaction mixture, and the mixture was extracted with ethyl acetate. The organic layer was washed with water and a 15 saturated aqueous solution of sodium chloride, followed by drying the organic layer over anhydrous sodium sulfate. The solvent was distilled off under reduced pressure, and the resulting crude product was purified by silica gel 20 column chromatography (chloroform) to obtain Compound 25 (161 mg, 80%) as a white solid.

[0115] [Equation 24]

mp 216-218°C. $^1\text{H-NMR}$ (CDCl_3) δ : 0.99-1.12 (28H, m), 2.12 (3H, s), 3.63 (1H, AB, J = 12 Hz), 3.70 (1H, AB, J = 13 Hz), 3.87 (1H, AB, J = 12 Hz), 4.11 (1H, d, J = 3 Hz), 4.13 (1H, AB, J = 13 Hz), 4.59 (1H, d, J = 3 Hz), 5.22 (1H, AB, J = 12 Hz), 5.29 (1H, AB, J = 12 Hz), 6.02 (1H, s), 7.31-7.53 (8H, m), 7.88 (1H, s), 8.30-8.33 (2H, m).

[0116] (25) Synthesis of Compound 26

In a stream of nitrogen, a 1M hexane solution of 5 boron trichloride (1.35 ml, 1.35 mmols) was added, while being cooled at -78°C, to a methylene chloride solution (7.5 ml) of Compound 25 (115 mg, 0.15 mmol), and the mixture was stirred for 1.5 hours, and further stirred for 10 2.5 hours under ice-cooled conditions. A saturated aqueous solution of sodium bicarbonate was added to the reaction mixture, and the mixture was extracted with ethyl acetate. The organic layer was washed with water and a saturated aqueous solution of sodium chloride, followed by drying the organic layer over anhydrous sodium sulfate. The solvent 15 was distilled off under reduced pressure, and the resulting crude product 26 (90 mg) was used for a next reaction without being purified.

[0117] [Equation 25]

$^1\text{H-NMR}$ (CDCl_3) δ : 0.99-1.13 (28H, m), 2.14 (3H, s), 2.57 (1H, AB, J = 13 Hz), 3.69 (1H, AB, J = 14 Hz), 3.70 (1H, AB, J = 13 Hz), 4.09 (1H, AB, J = 14 Hz), 4.11 (1H, d, J = 3 Hz), 4.42 (1H, d, J = 3 Hz), 6.22 (1H, s), 7.42-7.56 (3H, m), 7.91 (1H, s), 8.30-8.33 (2H, m).

20 [0118] (26) Synthesis of Compound 27

In a stream of nitrogen, phenoxyacetyl chloride (23 ml, 0.17 mmol) was added, under ice-cooled conditions, to a methylene chloride solution (1.5 ml) of Compound 26 (90 mg) and triethylamine (25 ml, 0.18 mmol), and the mixture was

stirred for 45 minutes. A saturated aqueous solution of sodium bicarbonate was added to the reaction mixture, and the mixture was extracted with ethyl acetate. The organic layer was washed with water and a saturated aqueous 5 solution of sodium chloride, followed by drying the organic layer over anhydrous sodium sulfate. The solvent was distilled off under reduced pressure, and the resulting crude product was purified by silica gel column chromatography (n-hexane:ethyl acetate = 2:1) to obtain 10 Compound 27 (49 mg, 42% from 23 in 2 steps) as a white solid.

[0119] [Equation 26]

mp 224-226°C. $^1\text{H-NMR}$ (CDCl_3) δ : 0.99-1.14 (28H, m), 2.13 (3H, s), 3.52 (1H, AB, J = 13 Hz), 3.75 (1H, AB, J = 13 Hz), 4.17 (1H, AB, J = 13 Hz), 4.18 (1H, d, J = 3 Hz), 4.23 (1H, AB, J = 13 Hz), 4.61 (1H, d, J = 3 Hz), 4.92 (1H, AB, J = 16 Hz), 5.05 (1H, AB, J = 16 Hz), 6.01 (1H, s), 6.95-7.02 (3H, m), 7.29 (2H, m), 7.42-7.57 (4H, m), 7.79 (1H, s), 8.30-8.33 (2H, m).

[0120] (27) Synthesis of Compound 28

15 In a stream of nitrogen, tetra-n-butylammonium fluoride (1M tetrahydrofuran solution, 0.12 ml, 0.12 mmol) was added, at room temperature, to a tetrahydrofuran solution (1.2 ml) of Compound 27 (45 mg, 0.06 mmol), and the mixture was stirred for 10 minutes. The solvent was 20 distilled off under reduced pressure, and the resulting crude product was purified by silica gel column chromatography (n-hexane:ethyl acetate = 1:1) to obtain Compound 28 (31 mg, 100%) as a white solid.

[0121] [Equation 27]

mp 148-150°C. $^1\text{H-NMR}$ (pyridine-d₅) δ : 1.94 (3H, s), 4.08 (1H, AB, J = 13 Hz), 4.23 (1H, AB, J = 12 Hz), 4.30 (1H, AB, J = 12 Hz), 4.73 (1H, AB, J = 13 Hz), 5.21 (1H, AB, J = 16 Hz), 5.28 (1H, d, J = 3 Hz), 5.42 (1H, AB, J = 16 Hz), 5.42 (1H, d, J = 3 Hz), 6.61 (1H, s), 6.95-7.00 (1H, m), 7.27-7.36 (4H, m), 7.45-7.54 (3H, m), 7.73 (1H, s), 8.54-8.56 (2H, m).

[0122] [Example 2] Synthesis and purification of oligonucleotide analogues

(1) Synthesis of 2',4'-BNA^{NC} modified
5 oligonucleotides

Oligonucleotide analogues (1) to (22) containing 2',4'-BNA^{NC} monomer units were synthesized by an automated nucleic acid synthesizer Expedite™ 8909 (ABI) on a 0.2 mmol scale in accordance with a standard phosphoroamidite protocol. The coupling time for the amidite and the 5'-terminal hydroxyl group was set at 94 seconds for the natural nucleoside amidite and at 300 seconds for the 2',4'-BNA^{NC} amidites (16, 17 and 21).

[0123] For the 2',4'-BNA^{NC} modified oligonucleotides having the 5'-terminal protected with a DMTr group and supported in a solid phase, removal (1.5 h) from a column using 28% aqueous ammonia was performed. The removed oligonucleotide analogues were reacted for 16 hours at 60°C in 28% aqueous ammonia for deprotection of all protective groups.

[0124] Simple purification by an NAP-10 column was carried out, and respective fractions obtained were subjected to UV analysis.

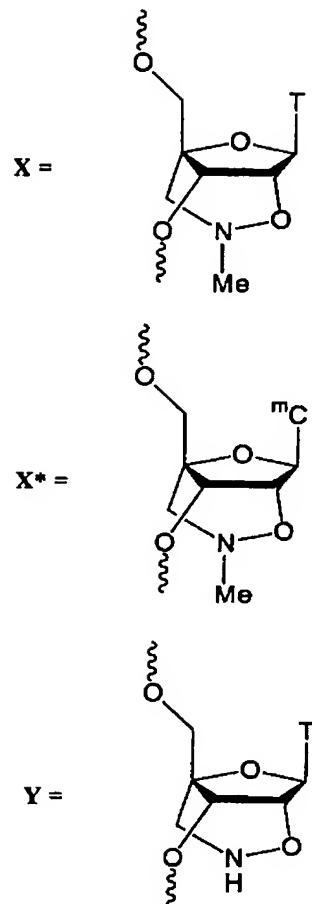
[0125] Based on the results of the UV measurements, 25 oligo-containing fractions were lyophilized, and the lyophilizates were further purified by reversed phase HPLC

(WakoPak^R WS-DNA column, 10.0 mm × 250 mm) (conditions: gradient elution with 8-16% acetonitrile for 30 min at a rate of 3 ml/min in 0.1M triethylammonium acetate buffer (pH 7.0)). The purity of the synthetic oligonucleotide analogues was confirmed by reversed phase HPLC (WakoPak^R WS-DNA column, 4.6 mm × 250 mm) (conditions: gradient elution with 8-16% acetonitrile for 30 min at a rate of 1 ml/min in 0.1M triethylammonium acetate buffer (pH 7.0)). The molecular weights were determined by MALDI-TOF-MASS measurement.

[0126] The synthetic oligonucleotide analogues (1) to (22) are indicated below.

[0127] [Chemical formula 7]

- ① 5'-GCGTTXTTGCT-3'
- ② 5'-GCGTTXTXTGCT-3'
- ③ 5'-GCGXTXTXTGCT-3'
- ④ 5'-GCGTTXXXTGCT-3'
- ⑤ 5'-GCGXXXXXXGCT-3'
- ⑥ 5'-TTTTT^mCTXT^mCT^mCT^mCT-3'
- ⑦ 5'-TTTTX^mCTXT^mCX^mCT^mCT-3'
- ⑧ 5'-TTTTT^mCXXX^mCT^mCT^mCT-3'
- ⑨ 5'-XTXTX^mCXTX^mCX^mCX^mCT-3'
- ⑩ 5'-TXTTX^mCTXT^mCX^mCTX^mT-3'
- ⑪ 5'-XTTTX^mCTTX^mCT^mCX^mCT-3'
- ⑫ 5'-TTTTTTTTXT-3'
- ⑬ 5'-GCGTTYTTTGCT-3'
- ⑭ 5'-GCGTTYTGTGCT-3'
- ⑮ 5'-GCGYTYTYTGCT-3'
- ⑯ 5'-GCGTTYYYYGCT-3'
- ⑰ 5'-GCGYYYYYYYGCT-3'
- ⑱ 5'-TTTTT^mCTYT^mCT^mCT^mCT-3'
- ⑲ 5'-TTTTY^mCTYT^mCY^mCT^mCT-3'
- ⑳ 5'-TTTTT^mCYYY^mCT^mCT^mCT-3'
- (21) 5'-YTYYTY^mCYYY^mCY^mCY^mCT-3'
- (22) 5'-TTTTTTTYT-3'



[0128] The amounts and yields of the 2',4'-BNA^{NC} modified oligonucleotides obtained are shown below.

[0129] [Table 1]

Oligonucleotide	Amount obtained (nmol)	Yield (%)
① 5'-GCGTTXTTGCT-3'	94.91	47.5
② 5'-GCGTTXTXTGCT-3'	118.67	59.3
③ 5'-GCGXTXTXTGCT-3'	126.12	63.1
④ 5'-GCGTTXXXTGCT-3'	66.94	33.5
⑤ 5'-GCGXXXXXXGCT-3'	99.40	49.7
⑥ 5'-TTTTT ^m CTXT ^m CT ^m CT ^m CT-3'	72.44	36.2
⑦ 5'-TTTX ^m CTXT ^m CX ^m CT ^m CT-3'	61.07	30.6
⑧ 5'-TTTTT ^m CXXX ^m CT ^m CT ^m CT-3'	60.40	30.2
⑨ 5'-XTXTX ^m CXTX ^m CX ^m CX ^m CT-3'	41.60	20.8
⑩ 5'-TXXTX ^m CTXT ^m CX ^m CTX ^m T-3'	25.40	12.7
⑪ 5'-XTTTX ^m CTTX ^m CT ^m CX ^m CT-3'	81.80	40.9
⑫ 5'-TTTTTTTTXT-3'	100.13	50.1

[0130] Furthermore, the results of the MALDI-TOF MS measurement of the synthetic 2',4'-BNA^{NC} modified oligonucleotides are tabulated below.

[0131] [Table 2]

Oligonucleotides	Calcd.For (M-H) ⁺	Found (M-H) ⁺
① 5'-GCGTTXTTGCT-3'	3689.47	3688.52
② 5'-GCGTTXTXTGCT-3'	3746.52	3746.89
③ 5'-GCGXTXTXTGCT-3'	3803.57	3804.93
④ 5'-GCGTTXXXTGCT-3'	3803.57	3803.18
⑤ 5'-GCGXXXXXXGCT-3'	3974.73	3975.31
⑥ 5'-TTTTT ^m CTXT ^m CT ^m CT ^m CT-3'	4553.11	4552.67
⑦ 5'-TTTTX ^m CTXT ^m CX ^m CT ^m CT-3'	4667.21	4667.36
⑧ 5'-TTTTT ^m CXXX ^m CT ^m CT ^m CT-3'	4667.21	4666.86
⑨ 5'-XTXTX ^m CXTX ^m CX ^m CX ^m CT-3'	4895.42	4896.42
⑩ 5'-TXTTX ^m CTXT ^m CX ^m CTX ^m T-3'	4781.32	4781.18
⑪ 5'-XTTTX ^m CTTX ^m CT ^m CX ^m CT-3'	4724.26	4723.73
⑫ 5'-TTTTTTTTXT-3'	3036.06	3036.36
⑬ 5'-GCGTTYTTGCT-3'	3675.44	3675.80
⑭ 5'-GCGTTYTGTGCT-3'	3718.47	3717.15
⑮ 5'-GCGYTYTYTGCT-3'	3761.49	3762.64
⑯ 5'-GCGTTYYYYTGCT-3'	3761.49	3761.24
⑰ 5'-GCGYYYYYYYGCT-3'	3890.57	3890.62
⑱ 5'-TTTTT ^m CTYT ^m CT ^m CT ^m CT-3'	4539.08	4540.11
⑲ 5'-TTTTY ^m CTYT ^m CY ^m CT ^m CT-3'	4625.13	4625.74
⑳ 5'-TTTTT ^m CYYY ^m CT ^m CT ^m CT-3'	4625.13	4625.71
(21) 5'-YTYT ^m CYYY ^m CY ^m CY ^m CT-3'	4797.23	4797.80
(22) 5'-TTTTTTTTYT-3'	3022.03	3021.93

[0132] [Experimental Example 1] Measurement of melting points (T_m) of 2',4'-BNA^{NC} modified oligonucleotides
 5 (evaluation of duplex-forming capacity)

The oligonucleotide analogues (1) to (5) and (13) to (17) synthesized in the Examples (i.e., antisense strands), and a sense strand of a natural DNA or RNA oligonucleotide were annealed. The melting points (T_m) of the annealing 10 products were measured to investigate the duplex-forming

capacity of the antisense strands.

[0133] Sample solutions (500 µL) with end concentrations of NaCl 100 mM, sodium phosphate buffer (pH 7.2) 10 mM, antisense strand 4 µM, and sense strand 4 µM were bathed in boiling water, and cooled down to room temperature over the course of 10 hours. A stream of nitrogen was flowed through a cell chamber of a spectrophotometer (Beckman, DU-650) to prevent dew formation, and the sample solutions were gradually cooled down to 5°C. Further, the sample solutions were maintained at 10°C for 20 minutes, and then their measurements were started. The temperature was raised up to 90°C at a rate of 0.5°C/min, and the ultraviolet absorption at 260 nm was measured.

[0134] The results are shown in Tables 3 and 4.

[0135] [Table 3]

Duplex-forming capacity with DNA strand (5'-AGCAAAAAACGC-3')
(Tm value evaluation): °C

Antisense strand	Tm	ΔTm/mod.
5'-GCGTTTTTGCT-3'	50	Natural
① 5'-GCGTTXTTTGCT-3'	49	-1
② 5'-GCGTTXTXTGCT-3'	49	-0.5
③ 5'-GCGXTXTXTGCT-3'	51	-0.3
④ 5'-GCGTTXXXTGCT-3'	50	0
⑤ 5'-GCGXXXXXXGCT-3'	61	+1.8
⑬ 5'-GCGTTYTTTGCT-3'	51	+1
⑭ 5'-GCGTTYTYTGCT-3'	52	+1
⑮ 5'-GCGYTYTYTGCT-3'	55	+1.7
⑯ 5'-GCGTYYYTGCT-3'	57	+2.3
⑰ 5'-GCGYYYYYYGCT-3'	73	+3.8

Conditions: 100 mM NaCl, 10 mM Na2HPO4 buffer (pH7.2), [strand] = 4 µM.

Tm was the average value of 3 or more experiments.

[0136] [Table 4]

Duplex-forming capacity with RNA strand (5'-AGCAAAAAACGC-3')
(Tm value evaluation): °C

Antisense strand	Tm	ΔTm/mod.
5'-GCGTTTTTGCT-3'	45	Natural
① 5'-GCGTTXTTGT-3'	50	+5
② 5'-GCGTTXTXTGCT-3'	56	+5.5
③ 5'-GCGXTXTXTGCT-3'	63	+6
④ 5'-GCGTXXXTGCT-3'	59	+4.7
⑤ 5'-GCGXXXXXXGCT-3'	80	+5.8
⑬ 5'-GCGTTYTTTGCT-3'	51	+6
⑭ 5'-GCGTTYTGTGCT-3'	56	+5.5
⑮ 5'-GCGYTYTGTGCT-3'	64	+6.3
⑯ 5'-GCGTTYYTGCT-3'	61	+5.3
⑰ 5'-GCGYYYYYGCT-3'	83	+6.3

Conditions: 100 mM NaCl, 10 mM Na₂HPO₄ buffer (pH7.2), [strand] = 4 μM.

Tm was the average value of 3 or more experiments.

[0137] Based on the above findings, the nucleotide analogues of the present invention have a much higher duplex-forming capacity with single-stranded RNA (sense strand) than their duplex-forming capacity with single-stranded DNA (sense strand), and are considered to be suitable for the antisense method.

[Experimental Example 2] Measurement of Tm of
2',4'-BNA^{NC} modified oligonucleotides (evaluation of
triplex-forming capacity)

In connection with the oligonucleotide analogues (6) to (11) and (18) to (21) synthesized in the Examples, the triplex-forming capacity with the target double-stranded DNA's indicated below was investigated by the same method as in Experimental Example 1. The salt concentrations and pH during measurement complied with the conditions described below the respective tables.

[0138] The results and experimental conditions are shown in Table 5.

20 [0139] [Table 5-1]

Table 5-1. T_m values of 2',4'-BNA^{NC} oligonucleotides with dsDNA.

target : 5' - d (GCTAAAAAGAAAGAGAGATCG) - 3' 3' - d (CGATTTTCTTTCTCTCTAGC) - 5'	
oligonucleotides	T_m (ΔT_m /modification) (°C)
5'-TTTTT ^m CTTT ^m CT ^m CT ^m CT-3'	33
⑥ 5'-TTTTT ^m CTXT ^m CT ^m CT ^m CT-3'	38 (+5.0)
⑦ 5'-TTTTX ^m CTXT ^m CX ^m CT ^m CT-3'	47 (+4.6)
⑧ 5'-TTTTT ^m CXXX ^m CT ^m CT ^m CT-3'	42 (+3.0)
⑨ 5'-XTXTX ^m CXTX ^m CX ^m CX ^m CT-3'	59 (+3.7)
⑩ 5'-TXTTX ^m CTXT ^m CX ^m CTX ^m T-3'	50 (+3.4)
⑪ 5'-XTTTX ^m CTTX ^m CT ^m CX ^m CT-3'	45 (+3.0)
⑯ 5'-TTTTT ^m CTYT ^m CT ^m CT ^m CT-3'	44 (+11.0)
⑯ 5'-TTTTY ^m CTYT ^m CY ^m CT ^m CT-3'	60 (+9.0)
㉐ 5'-TTTTT ^m CYYY ^m CT ^m CT ^m CT-3'	59 (+8.7)
(21) 5'-YTYTY ^m CYTY ^m CY ^m CY ^m CT-3'	78 (+6.4)

Conditions : 140 mM KCl, 7 mM Na₂HPO₄ buffer (pH 7.0), each strand 1.5 μM, 5°C
to 90°C (0.5°C/min). X*:2',4'-BNA^{NC}(N-Me)-^mC

[0140]

[Table 5-2]

Table 5-2. T_m values of 2',4'-BNA^{NC} oligonucleotides with dsDNA.

target : 5' - d (GCTGCTAAAAAGAAAGAGAGATCGTCG) - 3' 3' - d (CGACGATTTCCTCTCTAGCAGC) - 5'	
oligonucleotides	T_m (ΔT_m /modification) (°C)
5'-TTTTT ^m CTTT ^m CT ^m CT ^m CT-3'	43
⑥ 5'-TTTTT ^m CTXT ^m CT ^m CT ^m CT-3'	49 (+6.0)
⑦ 5'-TTTTX ^m CTXT ^m CX ^m CT ^m CT-3'	61 (+6.0)
⑧ 5'-TTTTT ^m CXXX ^m CT ^m CT ^m CT-3'	54 (+3.6)
⑨ 5'-XTXTX ^m CXTX ^m CX ^m CX ^m CT-3'	>80 (>+5)
⑩ 5'-XTTX ^m CTXT ^m CX ^m CTX ^m T-3'	63 (+4.0)
⑪ 5'-XTTTX ^m CTTX ^m CT ^m CX ^m CT-3'	59 (+4.0)
⑫ 5'-TTTTT ^m CTYT ^m CT ^m CT ^m CT-3'	55 (+12.0)
⑬ 5'-TTTTY ^m CTYT ^m CY ^m CT ^m CT-3'	73 (+10.0)
⑭ 5'-TTTTT ^m CYYY ^m CT ^m CT ^m CT-3'	71 (+9.3)
(21) 5'-YTYYT ^m CYTY ^m CY ^m CY ^m CT-3'	>80 (>+5)

Conditions : 140 mM KCl, 10 mM MgCl₂, 7 mM Na₂HPO₄ buffer (pH 7.0), each strand 1.5 μM, 5°C to 90°C (0.5°C/min). X: 2',4'-BNA^{NC}(N-Me)-^mC

[0141] [Table 5-3]

Table 5-3. Sequence selective triplex-formation of 2',4'-BNA^{NC} oligonucleotides.

Z	M-N-	T_m ($\Delta T_m = T_{m(mismatch)} - T_{m(match)}$) (°C)			
		A-T	G-C	C-G	T-A
natural-T		43	21 (-22)	25 (-18)	18 (-25)
⑥ X [2',4'-BNA ^{NC} (N-Me)]		50	24 (-26)	22 (-28)	14 (-36)
⑭ Y [2',4'-BNA ^{NC} (N-H)]		55	30 (-25)	28 (-27)	25 (-30)

Conditions: 140 mM KCl, 7 mM sodium phosphate buffer (pH 7.0), each strand 1.5 μM, 5°C to 90°C (0.5°C/min). C: 2'-deoxy-5-methylcytidine.

[0142] As shown in Table 5, the oligonucleotide analogues (6) to (11) and (18) to (21) of the present

invention were found to have an excellent triplex-forming capacity. Their sequence selectivity was also superior to that of the natural antisense strand. Thus, they are believed to be very useful for the antigene method as well.

5 Experimental Example 3: Measurement of enzyme
resistance

Natural (DNA oligonucleotide) and nonnatural oligonucleotides shown below were examined for resistance to exonuclease which degrades the oligonucleotide from the 10 3'-side.

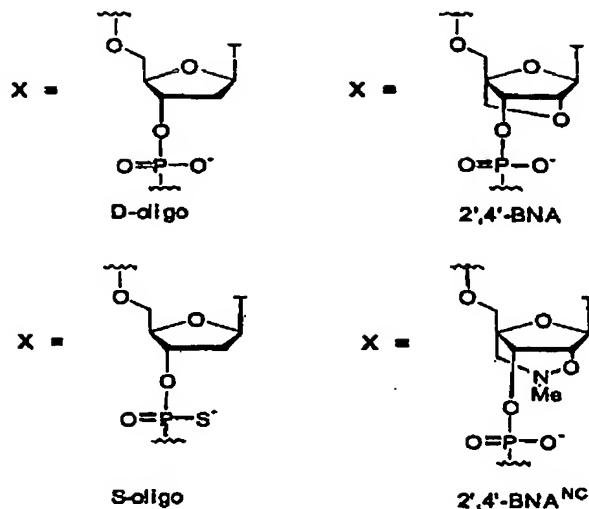
[0143] 1) Each oligonucleotide and snake venom phosphodiesterase (Boehringer Mannheim) as 3'-exonuclease were added, at concentrations of 25 mg/mL and 0.5 mg/mL, into 400 mL of 50 mM Tris-HCl buffer (pH 8.0) containing 15 10 mM MgCl₂, and the mixture was held at 37°C.

[0144] 2) After a constant period of time, the survival rate of each oligonucleotide was measured by HPLC.

[0145] The sequences of the oligonucleotides used in the measurement are shown below.

20 [0146] 5'-TTTTTTTXXT-3'

[0147] [Chemical Formula 8]



[0148] where when X is a DNA monomer, the oligonucleotide is a completely natural DNA oligonucleotide, and the oligonucleotide containing other nucleotide 5 analogue is a partially nonnatural oligonucleotide; moreover, the oligonucleotide analogue with X being 2',4'-BNA^{NC}(N-Me) is the oligonucleotide analogue of the present invention.

Changes over time in the survival rates of the 10 respective oligonucleotides, measured by HPLC, are shown in Table 6 and Fig. 1.

[0149] In Table 6 and Fig. 1, "% of intact oligonucleotide" refers to the survival rate (%) (HPLC determination) of the undegraded oligonucleotide at 15 measurement time points to the undegraded oligonucleotide at 0 time point.

[0150] [Table 6]

Evaluation of enzyme resistance

oligonucleotide	% of intact oligonucleotide					
	0min	5min	10min	20min	40min	90min
D-oligo	100	0	0	0	0	0
2',4'-BNA	100	37	24	5	1	0
S-oligo	100	94	92	91	83	70
2',4'-BNA ^{NC}	100	100	99	98	94	80

[0151] The results of Table 6 and Fig. 1 show that the
5 oligonucleotide analogue of the present invention had
excellent enzyme resistance as compared with natural and
other nonnatural oligonucleotides.